

# Today's outline - October 05, 2021



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- Information about:

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- Information about:
  - (a) Final presentation

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- Information about:
  - (a) Final presentation
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- SAXS papers

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Reading Assignment: Chapter 5.2–5.3



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Reading Assignment: Chapter 5.2–5.3

Homework Assignment #04:

Chapter 4: 2,4,6,7,10

due Tuesday, October 19, 2021

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due Tuesday, October 19, 2021

Homework Assignment #05:

Chapter 5: 1,3,7,9,10

due Tuesday, November 02, 2021



1. Choose paper for presentation



1. Choose paper for presentation
2. Clear it with me!



1. Choose paper for presentation
2. Clear it with me!
3. Do some background research on the technique



1. Choose paper for presentation
2. Clear it with me!
3. Do some background research on the technique
4. Prepare a 15 minute presentation



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3. Do some background research on the technique
4. Prepare a 15 minute presentation
5. Be ready for questions!



1. Come up with a potential experiment





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2. Make sure it is a different technique than your final presentation

# Final project



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4. Find appropriate beamline(s) and if needed contact the beamline scientists (they are used to it)



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5. Lay out proposed experiment (you can ask for help!)



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4. Find appropriate beamline(s) and if needed contact the beamline scientists (they are used to it)
5. Lay out proposed experiment (you can ask for help!)
6. Make sure to give reasonable answers for all the questions
7. Put me as one of the investigators of the proposal



The SAXS scattered intensity from a dilute solution depends on the single particle form factor,  $\mathcal{F}(\vec{Q})$ , the volume of the particle,  $V_p$ , and the density difference from the solvent,  $\Delta\rho = (\rho_{sl,p} - \rho_{sl,0})$



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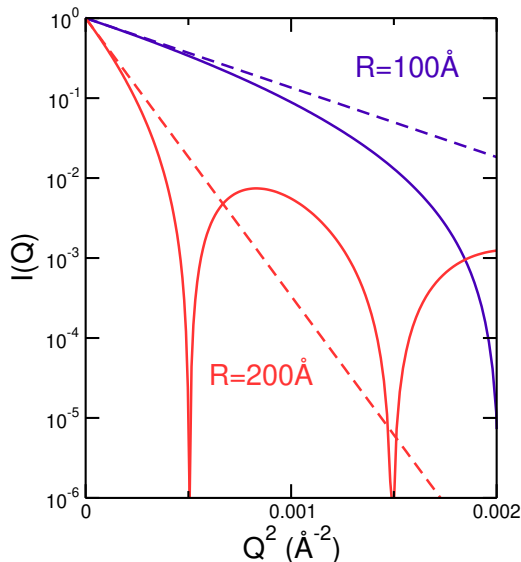


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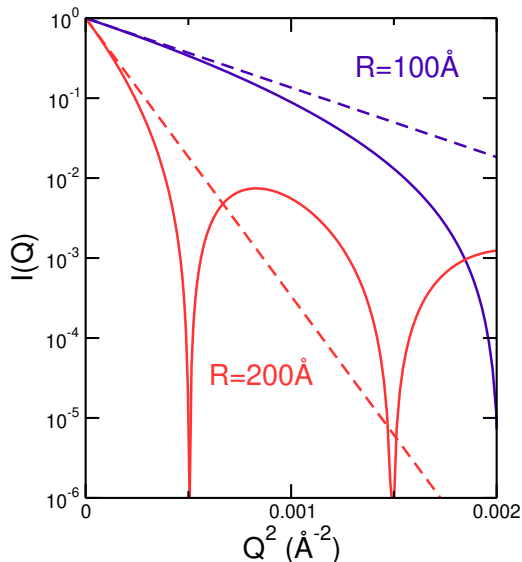
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$$R_g^2 = \frac{\int_{V_p} \rho_{sl,p}(\vec{r}) r^2 dV_p}{\int_{V_p} \rho_{sl,p}(\vec{r}) dV_p}$$



# Inter-particle interactions



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- SAXS of irradiated Zn nanoparticles
- Nucleation and growth of & glycine crystals

# SAXS of irradiated Zn nanoparticles



Zn nanoparticles formed in SiO<sub>2</sub> by ion implantation irradiated with high energy Xe<sup>+14</sup> ions.

"Shape elongation of embedded Zn nanoparticles induced by swift heavy ion irradiation: A SAXS study", H. Amekura, K. Kono, N. Okubo, and N. Ishikawa, *Phys. Status Solidi B* 252, 165-169 (2015).

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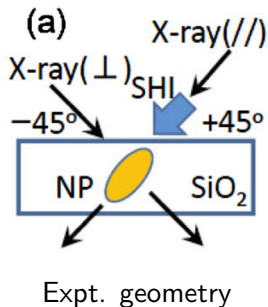
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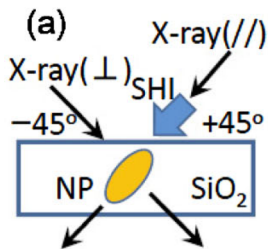
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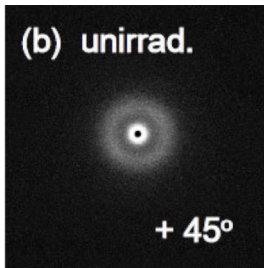


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Expt. geometry



Unirradiated

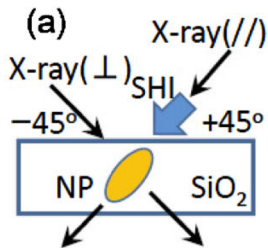
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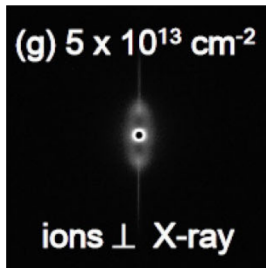


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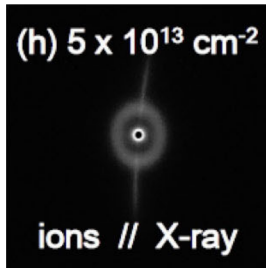
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Irradiated  $\perp$  x-rays



Irradiated  $\parallel$  x-rays

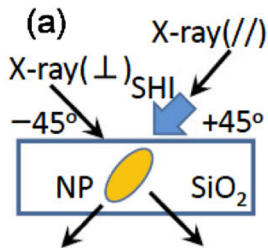
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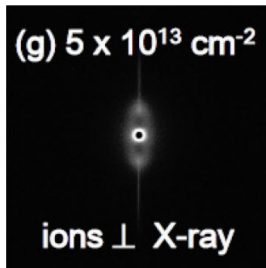


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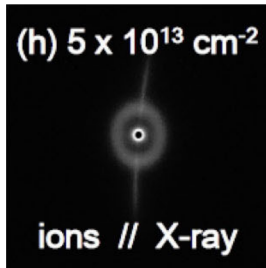
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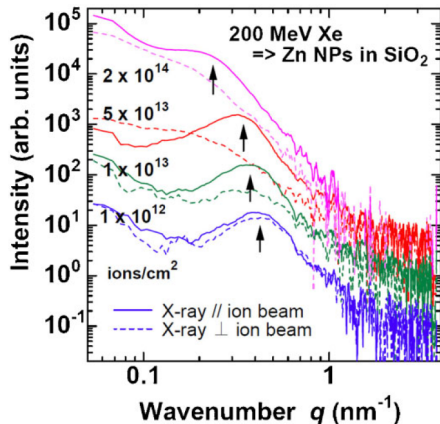
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Straight lines from ion tracks, seen in both directions and which persist to the highest fluences.

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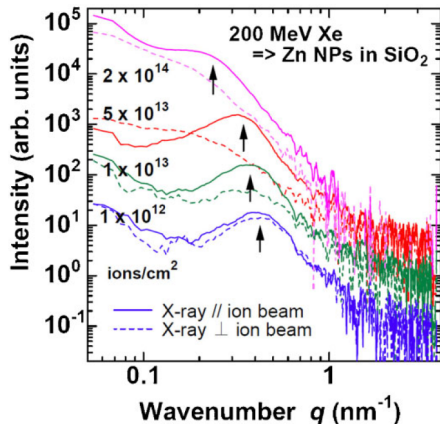
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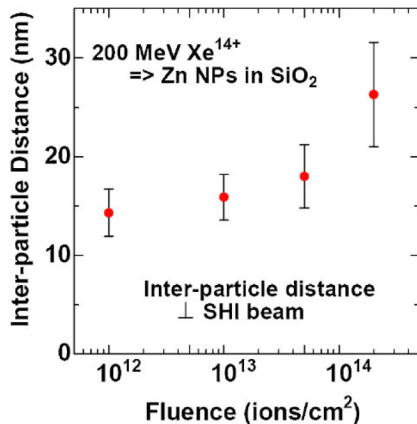
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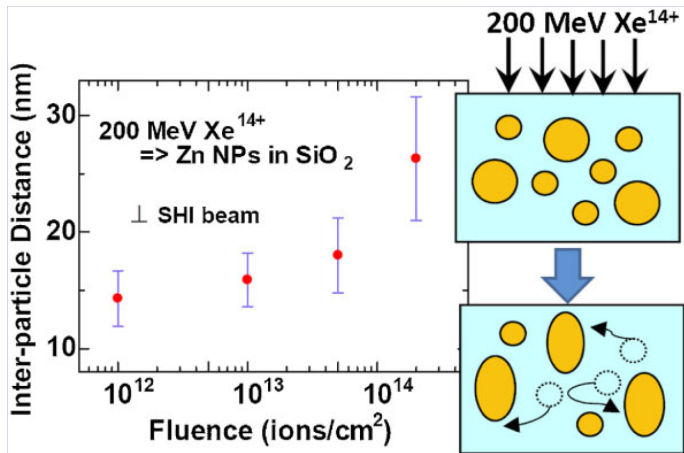
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Interparticle distance increases as a function of irradiation fluence

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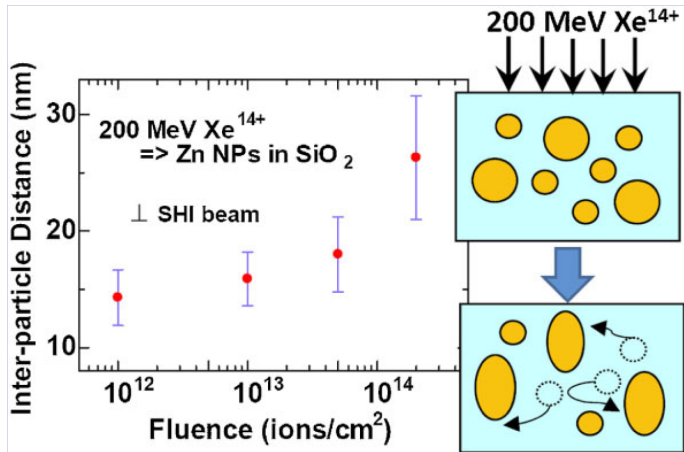


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Growth of interparticle spacing is due to dissolution and re-agglomeration with fluence leading to larger interparticle spacings



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# Nucleation & growth of glycine



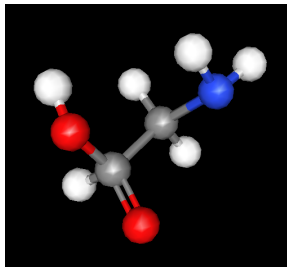
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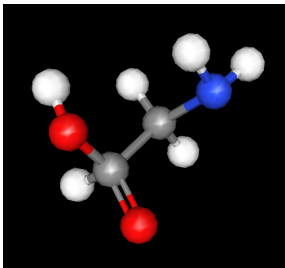


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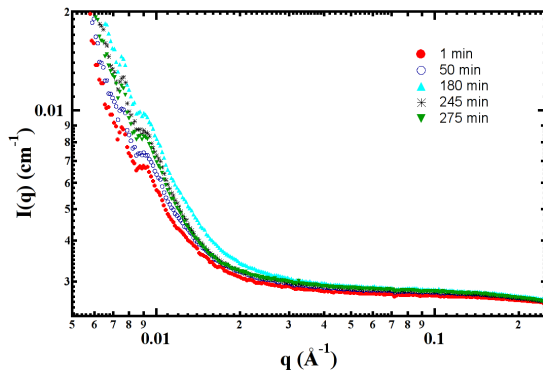
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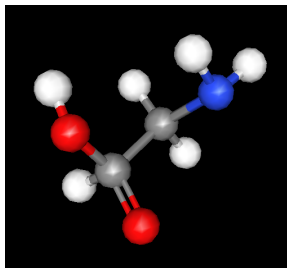


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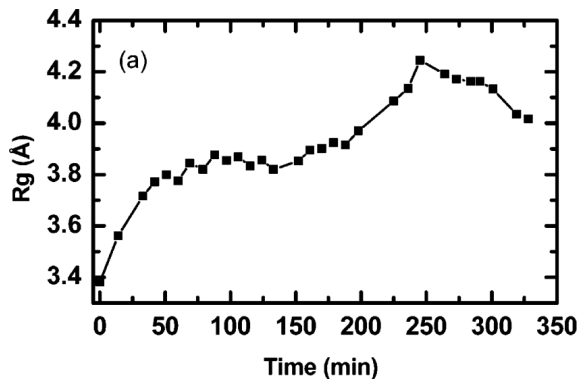
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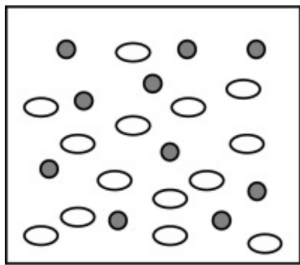
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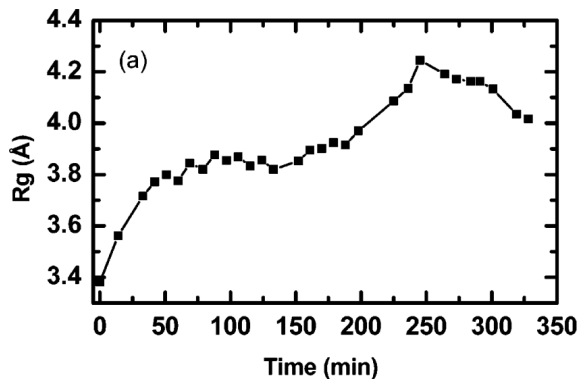


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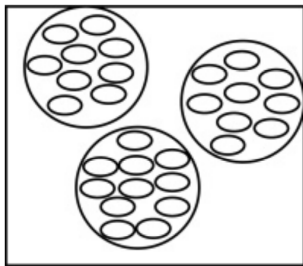


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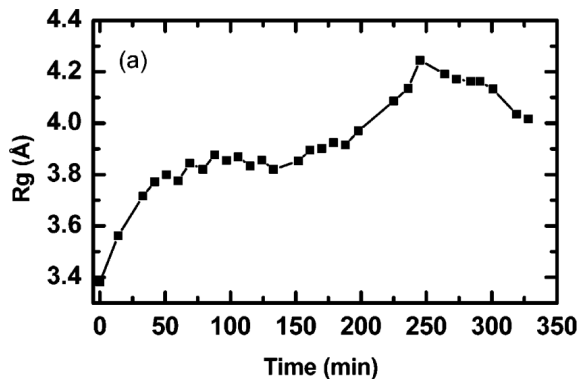


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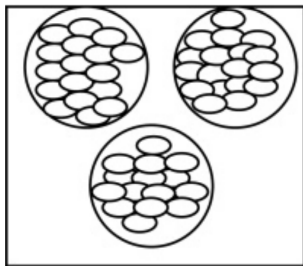


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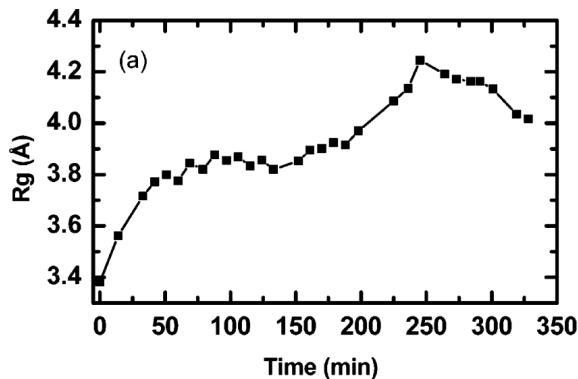
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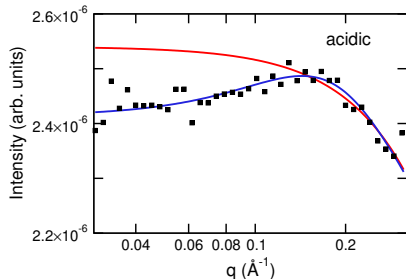
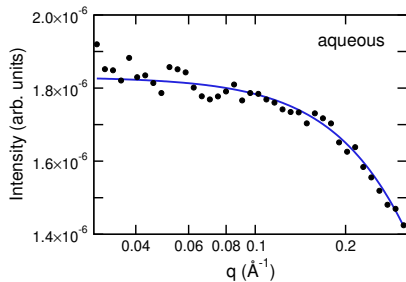


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Different polymorphs are produced under varying conditions so study extended to both neutral and acidic solutions

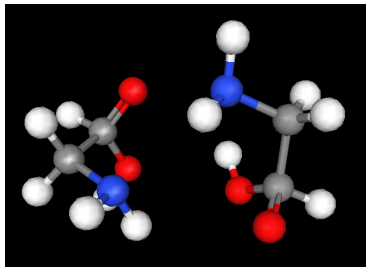
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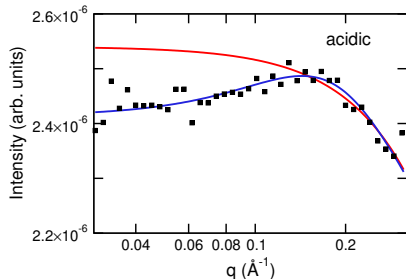
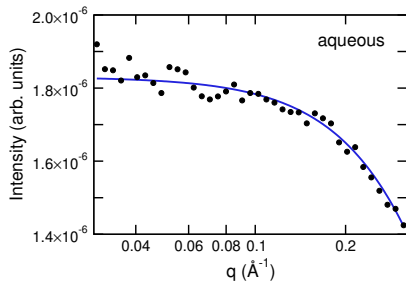
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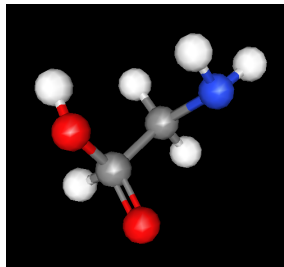
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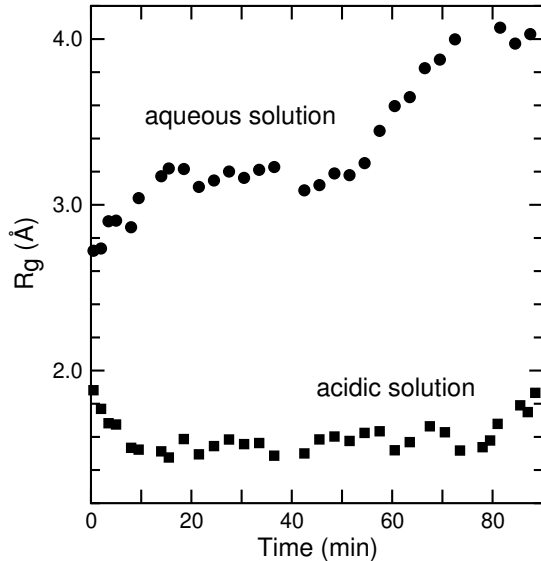
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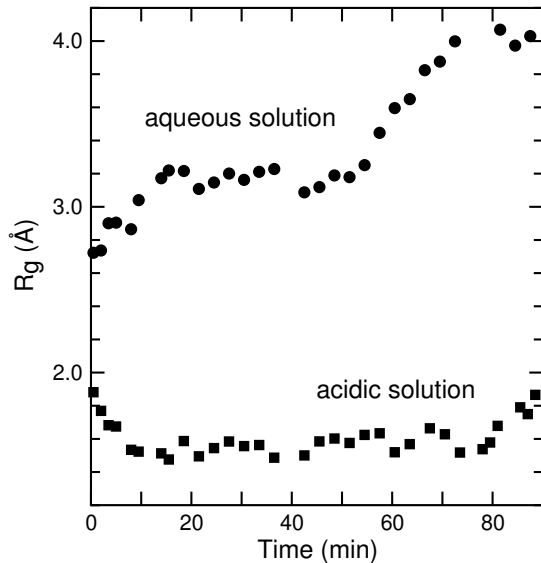






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In acidic solution,  $R_g$  remains small and implies that no dimerization or aggregation occurs before nucleation

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# Size exclusion chromatography SAXS



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Obtaining information about  $R_g$  and the Porod region, combined with modeling and the known crystallographic structures can give a more complete picture of how these molecules function.



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A major problem in these systems is aggregation and impurities. Pre-purification of samples is important but if they are left for some time before the SAXS measurement is performed, there can be decomposition.



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Even without any aggregation or decomposition, separation into a monodisperse molecule size is challenging.

# Size exclusion chromatography SAXS



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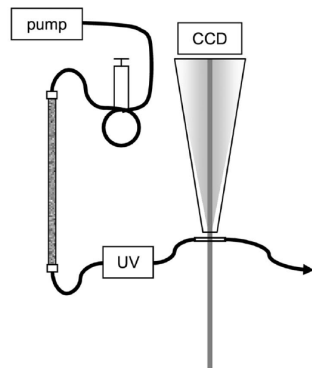
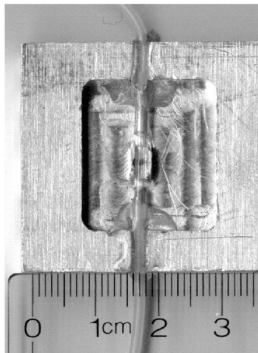
Obtaining information about  $R_g$  and the Porod region, combined with modeling and the known crystallographic structures can give a more complete picture of how these molecules function.

A major problem in these systems is aggregation and impurities. Pre-purification of samples is important but if they are left for some time before the SAXS measurement is performed, there can be decomposition.

Even without any aggregation or decomposition, separation into a monodisperse molecule size is challenging.

Mathew, Mirza & Menhart, "Liquid-chromatography-coupled SAXS for accurate sizing of aggregating proteins," *J. Synchrotron Rad.* **11**, 314-318 (2004) developed a technique which is now being used routinely in biological SAXS, called Size Exclusion Chromatography SAXS.

# Size exclusion chromatography SAXS



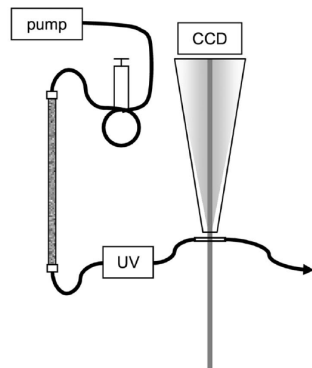
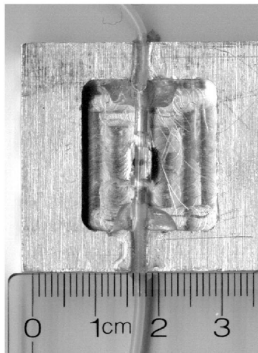
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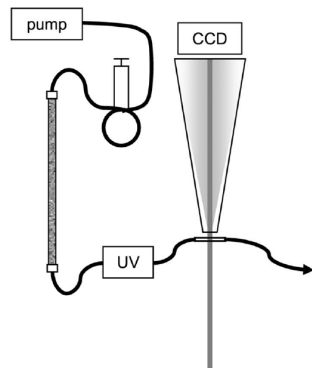
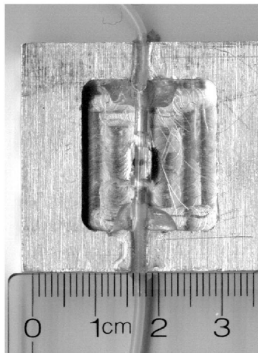
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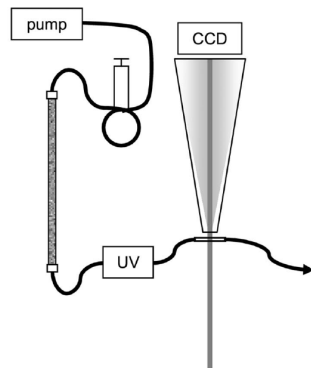
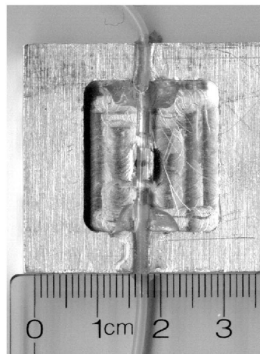
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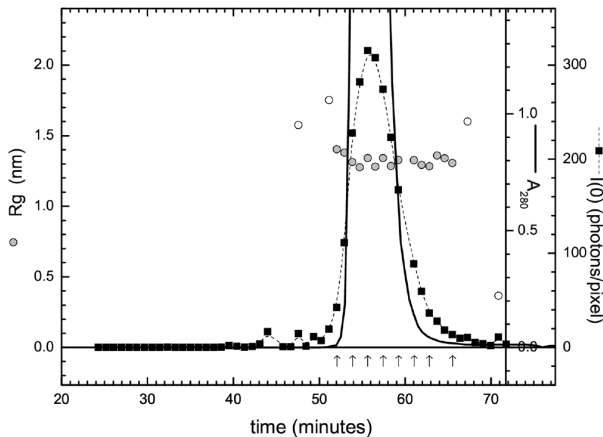


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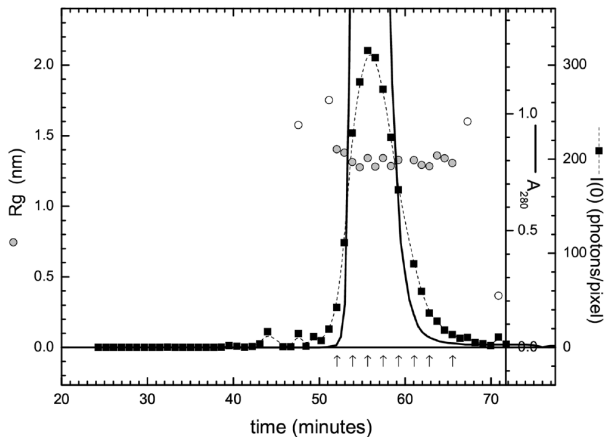


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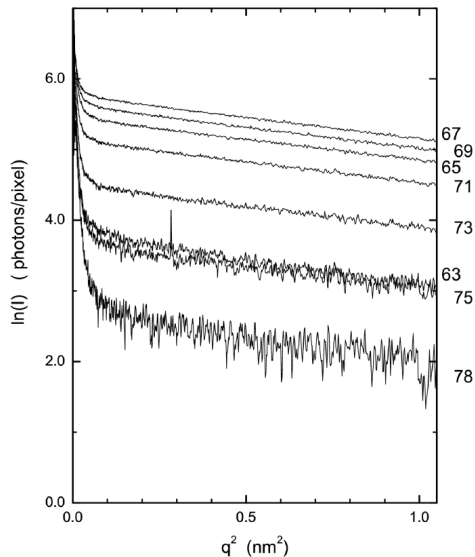
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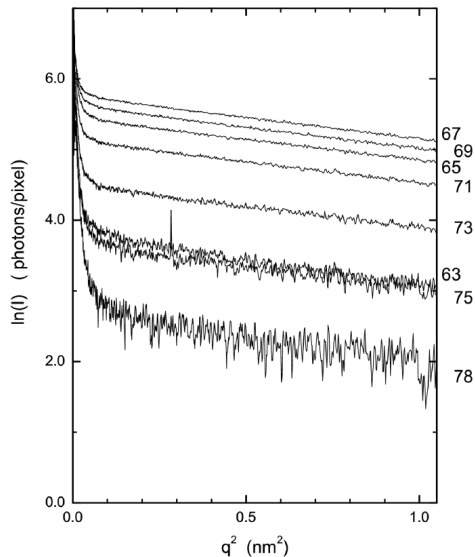
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# Cytochrome c - Guinier plots



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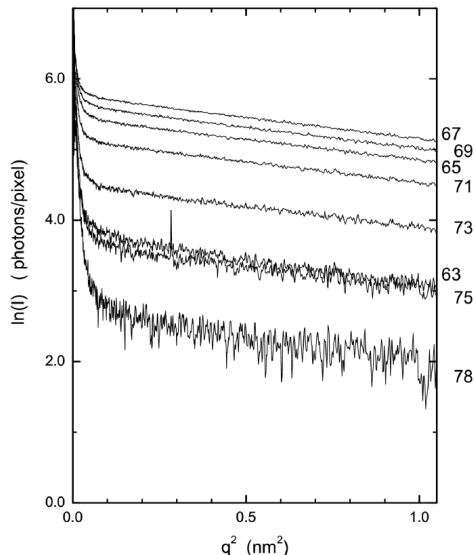
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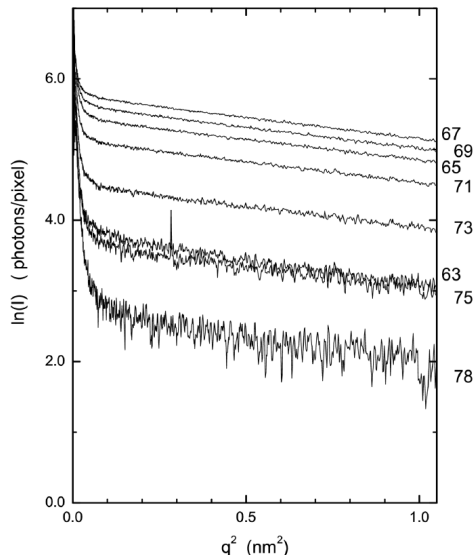
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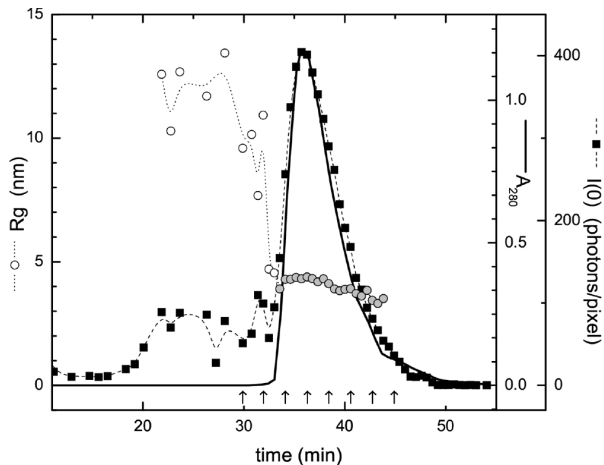
Even lowest intensity data set gives a consistent  $R_g$ .

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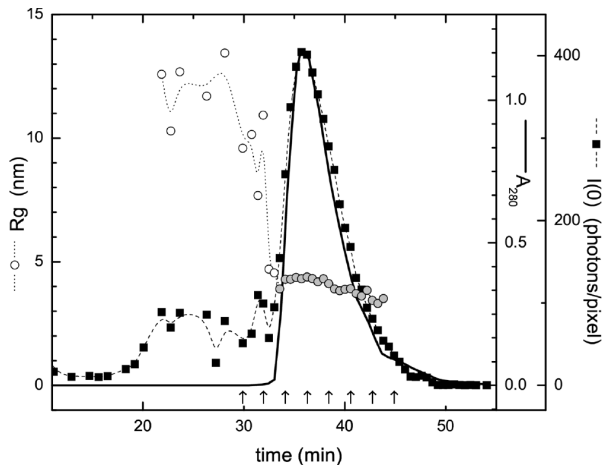


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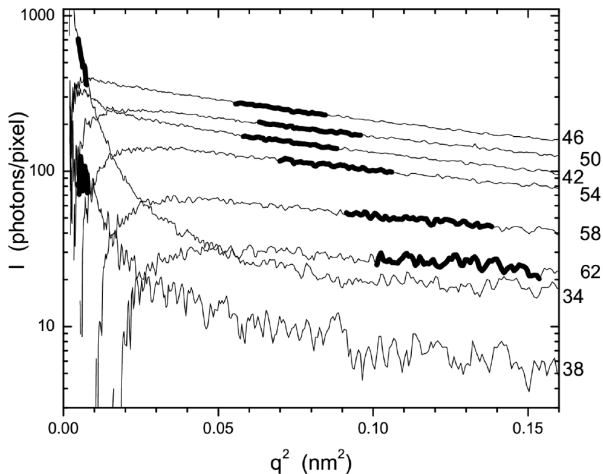
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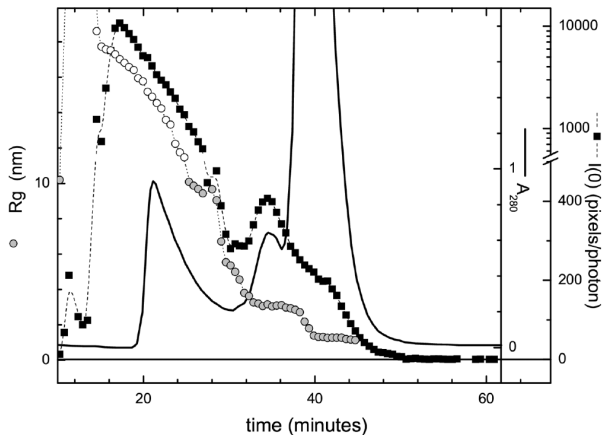
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The three components show consistent  $R_g$  and can be individually identified despite the overlap.



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# Porosity in CaO calcination



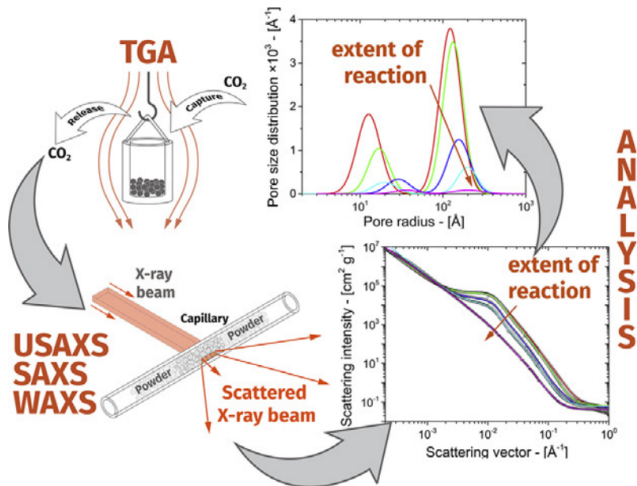
SAXS was used to study the nature of the porosity and particle sizes of CaO obtained by calcining  $\text{CaCO}_3$ .

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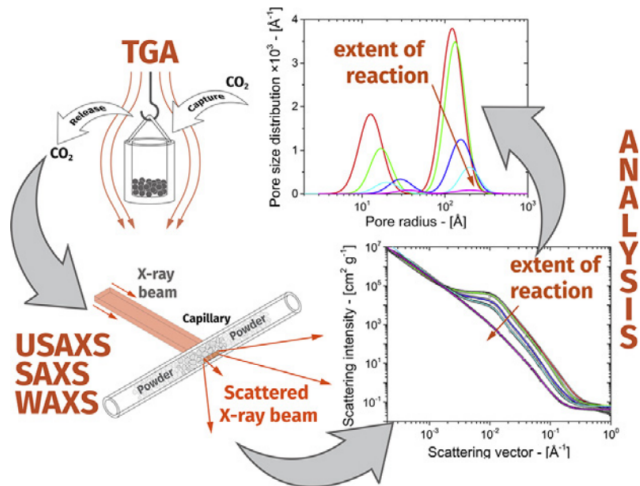
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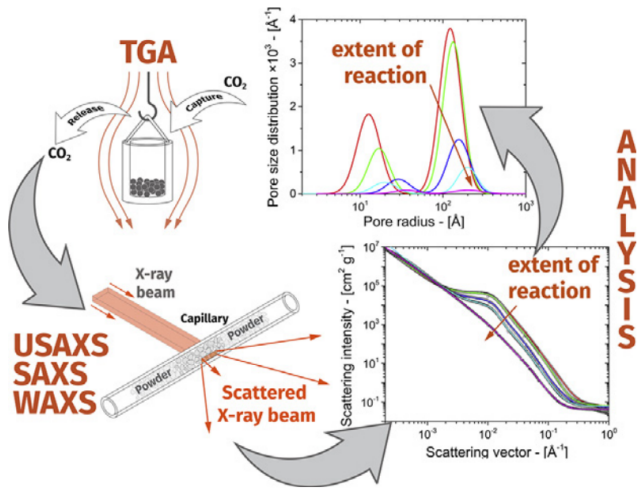
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The samples were studied ex-situ at Sector 9-ID using USAXS and analyzed with a unified fit model

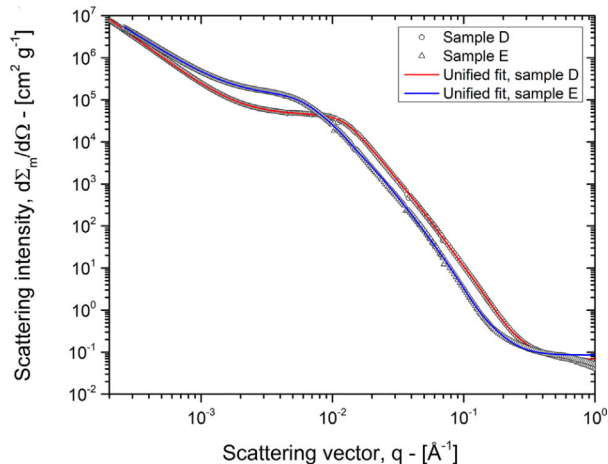


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Sample D was calcined at 900 °C for 50 minutes while sample E was calcined at the same temperature for 240 minutes



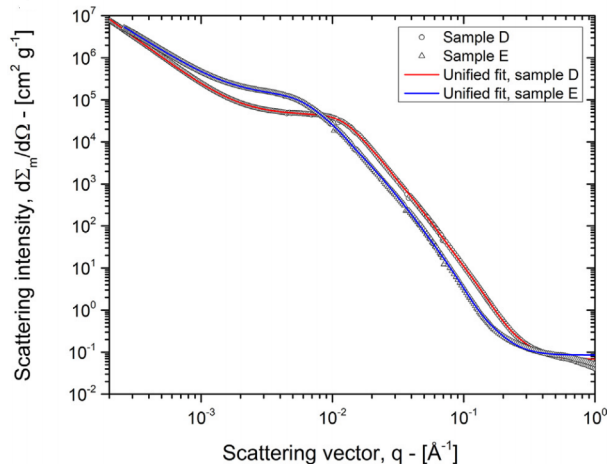
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The SAXS shows the grain growth evolution between the two samples and it is clear that the samples need a multilevel unified fit

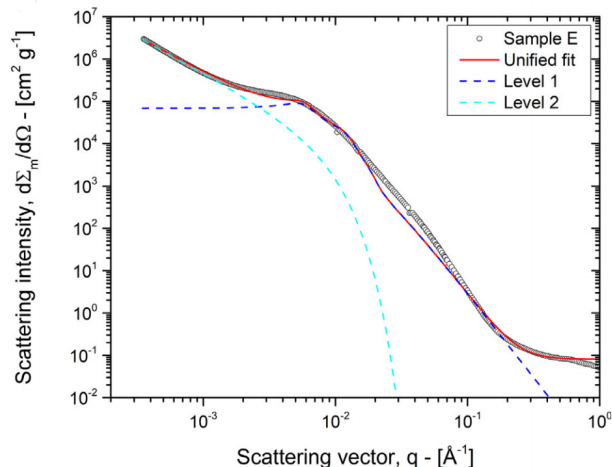


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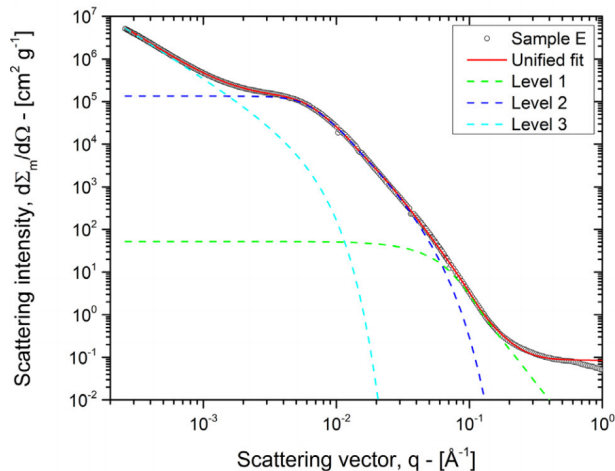
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A three level fit works well for the calcined samples and from this one can extract the pore sizes for two different pore populations in the calcined samples

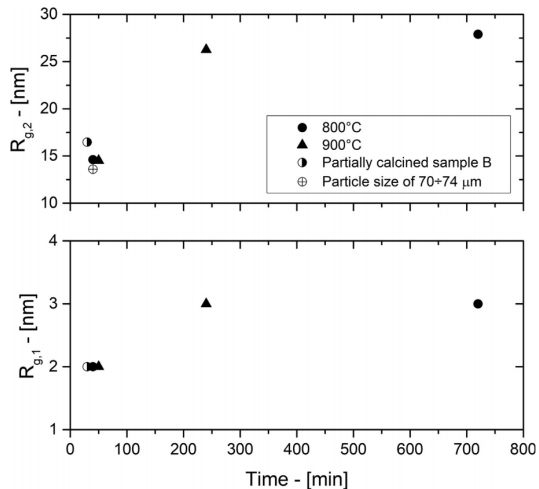


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Fitting a series of samples calcined at varying temperatures and times shows the evolution of the radii of gyration of the two populations corresponding to the pore sizes



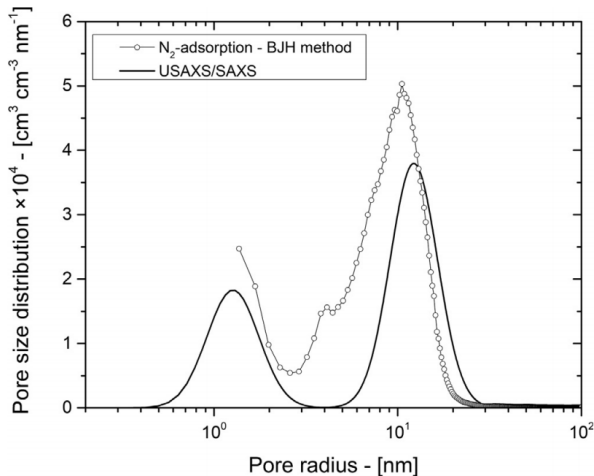
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The resulting pore size distributions correspond well to those measured using gas adsorption methods



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We will now proceed to develop a model for this kind of scattering starting with some definitions in 2D space.

# Types of lattice vectors

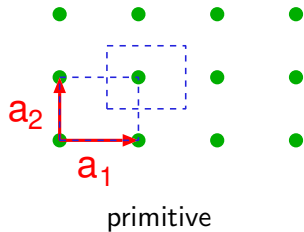


$$\vec{R}_n = n_1 \vec{a}_1 + n_2 \vec{a}_2$$

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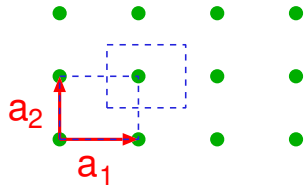
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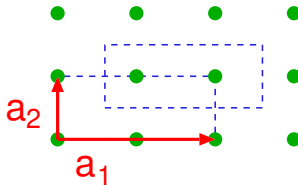
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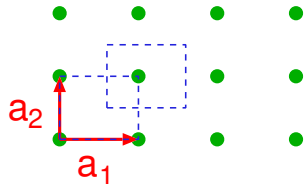


non-primitive

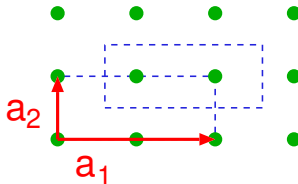
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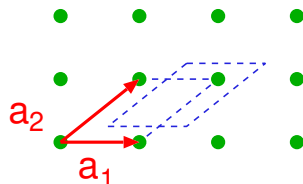
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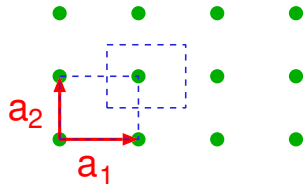


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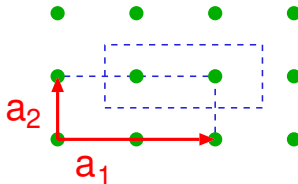
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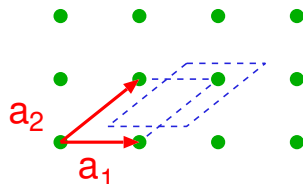
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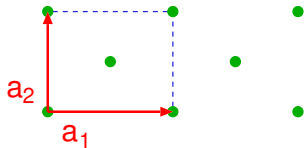
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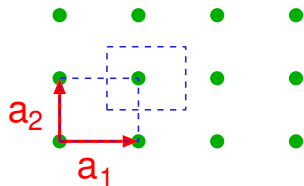
sometimes conventional axes...



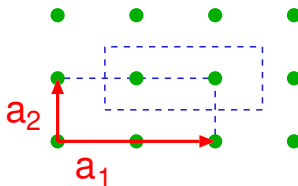
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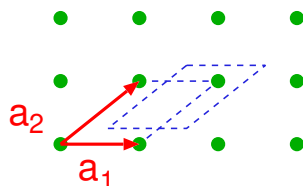
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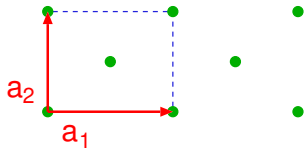
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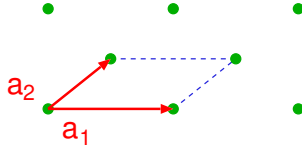
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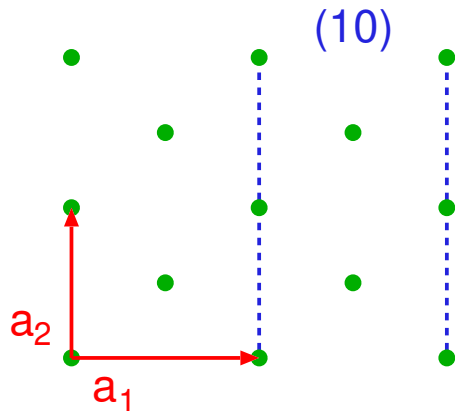


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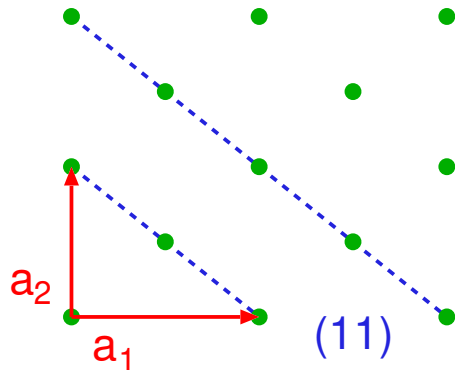
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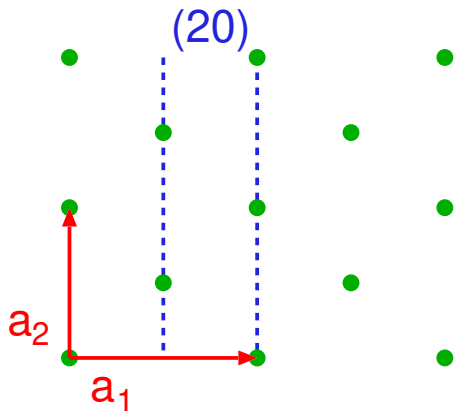
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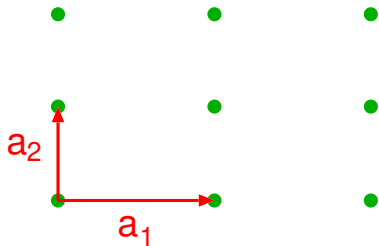
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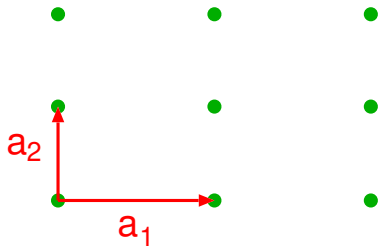
for a lattice with orthogonal unit vectors

$$\frac{1}{d_{hk}^2} = \frac{h^2}{a_1^2} + \frac{k^2}{a_2^2}$$

# Reciprocal lattice



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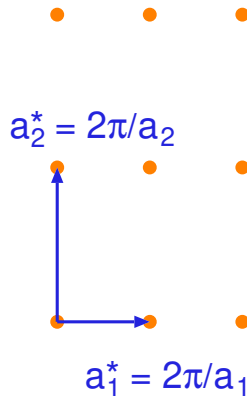
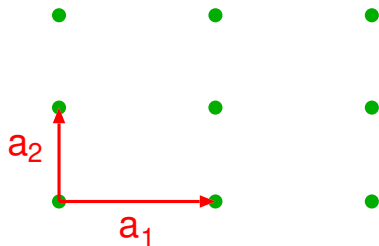


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the lattice, which is a collection of points in space, can be written

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$$\mathcal{C}(x) = \sum_n \mathcal{B}(x - na)$$

the lattice, which is a collection of points in space, can be written

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$$F^{crystal}(\vec{Q}) = \sum_I^N f_I(\vec{Q}) e^{i\vec{Q} \cdot \vec{r}_I}$$

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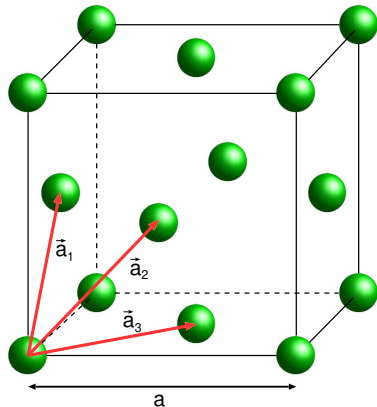
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# The FCC reciprocal lattice



The primitive lattice vectors of the face-centered cubic lattice are

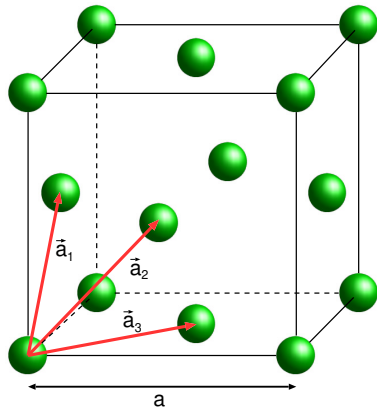


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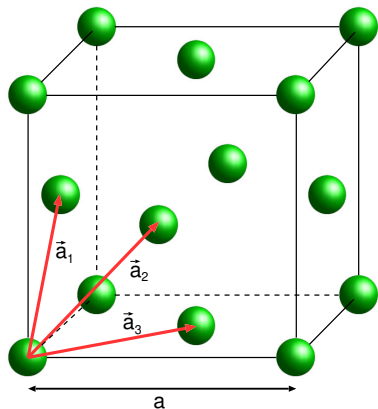


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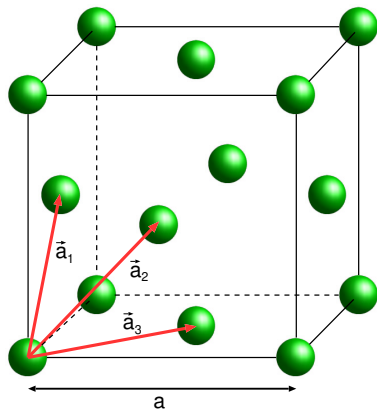


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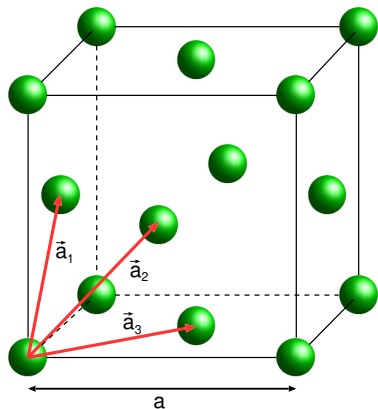


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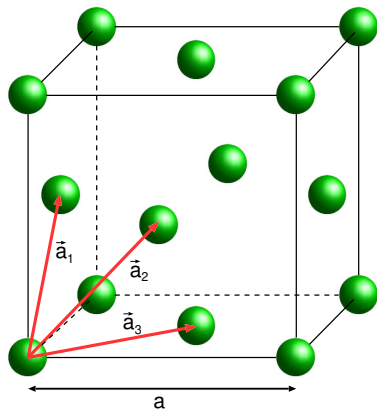
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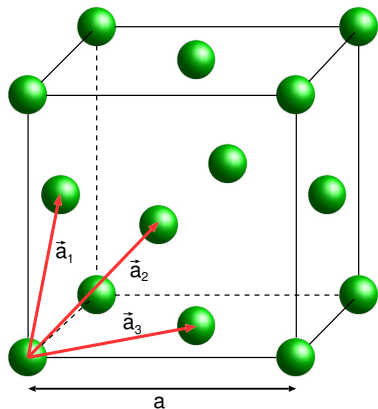
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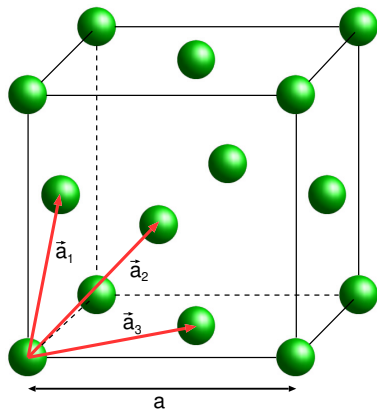
$$v_c = \vec{a}_1 \cdot \vec{a}_2 \times \vec{a}_3 = \vec{a}_1 \cdot \frac{a^2}{4} (\hat{y} + \hat{z} - \hat{x})$$

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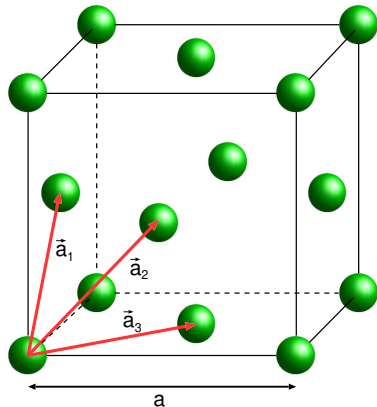
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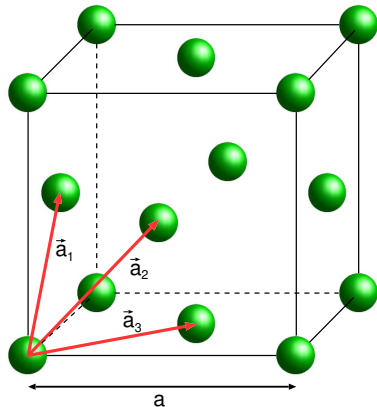
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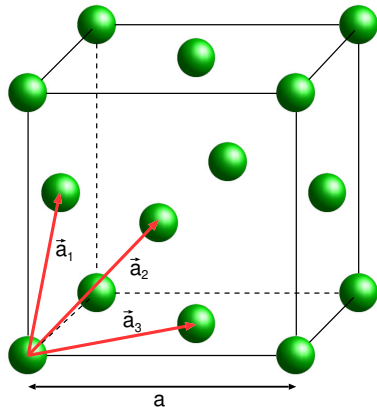
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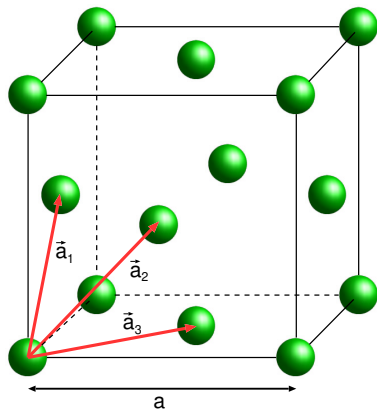
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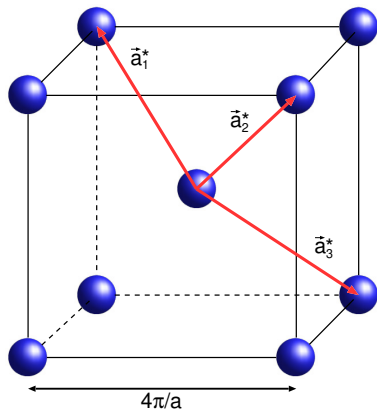


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since  $\delta(a^*\xi) = \delta(\xi)/a^*$

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