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• Scattering review

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Reading Assignment: Chapter 4.3–4.4

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Homework Assignment #03: Chapter 3: 1,3,4,6,8 due Tuesday, October 05, 2021

- Scattering review
- Kinematical scattering
- Liquid scattering

Reading Assignment: Chapter 4.3–4.4

Homework Assignment #03: Chapter 3: 1,3,4,6,8 due Tuesday, October 05, 2021 Homework Assignment #04: Chapter 4: 2,4,6,7.10 due Tuesday, October 19, 2021



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Consider systems where there is only weak scattering, with no multiple scattering effects. We begin with the scattering of x-rays from two electrons.



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The scattering from the second electron will have a phase shift of $\phi = \vec{Q} \cdot \vec{r}$.



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for many electrons

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for many electrons





for many electrons



generalizing to a crystal



for many electrons

generalizing to a crystal





for many electrons

generalizing to a crystal

 $\begin{aligned} \mathcal{A}(\vec{Q}) &= -r_0 \sum_{j} e^{i\vec{Q}\cdot\vec{r_j}} \\ \mathcal{A}(\vec{Q}) &= -r_0 \sum_{N} e^{i\vec{Q}\cdot\vec{R_N}} \sum_{j} e^{i\vec{Q}\cdot\vec{r_j}} \end{aligned}$

Since experiments measure $I \propto A^2$, the phase information is lost. This is a problem if we don't know the specific orientation of the scattering system relative to the x-ray beam.



for many electrons

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Since experiments measure $I \propto A^2$, the phase information is lost. This is a problem if we don't know the specific orientation of the scattering system relative to the x-ray beam.

We will now look at the consequences of this orientation and generalize to more than two electrons

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Two electrons — fixed orientation

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The expression

$$I(\vec{Q}) = 2r_0^2 \left(1 + \cos(\vec{Q}\cdot\vec{r})\right)$$

assumes that the two electrons have a specific, fixed orientation. In this case the intensity as a function of Q is.

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Fixed orientation is not the usual case, particularly for solution and small-angle scattering.





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$$Z_{He} = 2 \qquad Z_{Ar} = 18$$



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As an example take the CF_4 molecule



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The plot shows the structure factor of CF_4 ,

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The plot shows the structure factor of CF_4 , its orientationally averaged structure factor, and the form factor factor of Mo which has the same number of electrons as CF_4

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The logarithmic plot shows the spherically averaged structure factor compared to the inelastic scattering for CF_4

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Ordered 2D crystal

Amorphous solid or liquid



Ordered 2D crystal



Amorphous solid or liquid





Ordered 2D crystal



Amorphous solid or liquid



Take a circle (sphere) of radius r and thickness dr and count the number of atom centers lying within the ring.

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Take a circle (sphere) of radius r and thickness dr and count the number of atom centers lying within the ring. Then expand the ring radius by dr to map out the radial distribution function g(r)

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Total scattered intensity



Consider a mono-atomic (-molecular) system where the total scattered intensity is given by

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+ $f(\vec{Q})^{2} \rho_{at} \sum_{n} \int_{V} e^{i\vec{Q}\cdot(\vec{r}_{n}-\vec{r}_{m})} dV_{m} = I^{SRO}(\vec{Q})$

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$\mathsf{S}(\mathsf{Q})$ - the liquid structure factor

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Which is the sine Fourier Transform of the deviation of the atomic density from its average, $\mathcal{H}(r) = 4\pi r [g(r) - 1]$

Carlo Segre (Illinois Tech)



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The relation between radial distribution function and structure factor can be extended to multi-component systems where $g(r) \rightarrow g_{ij}(r)$ and $S(Q) \rightarrow S_{ij}(Q)$.

Liquid Ni metal was suspended electrostatically and allowed to cool from its liquidus temperature of 1450° C.

"Difference in Icosahedral Short-Range Order in Early and Late Transition Metal Liquids," G.W. Lee et al. *Phys. Rev. Lett* **93**, 037802 (2004).

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Details in the shape of the oscillations can be indicative of distortions in the icosahedra which depend on the metal species.

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