

Today's Outline - October 12, 2016

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- SAXS review

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Reading assignment: Chapter 5.2

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Homework Assignment #04:

Chapter 4: 2, 4, 6, 7, 10

due Monday, October 24, 2016

SAXS review

The SAXS scattered intensity from a dilute solution depends on the single particle form factor, $\mathcal{F}(\vec{Q})$, the volume of the particle, V_p , and the density difference from the solvent, $\Delta\rho = (\rho_{sl,p} - \rho_{sl,0})$

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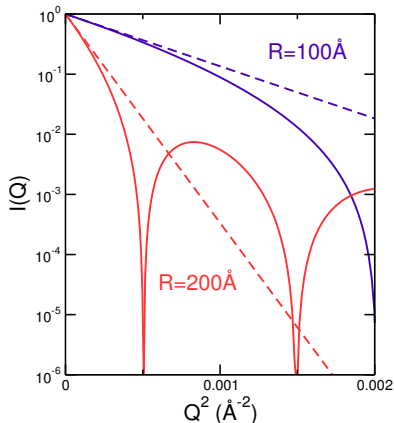
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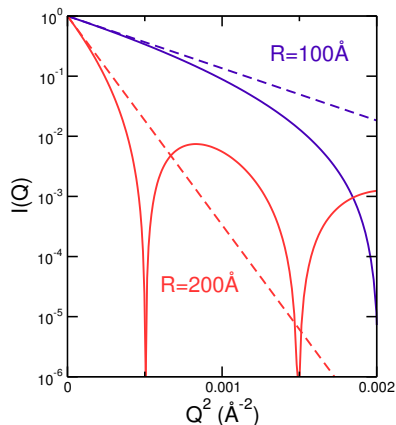
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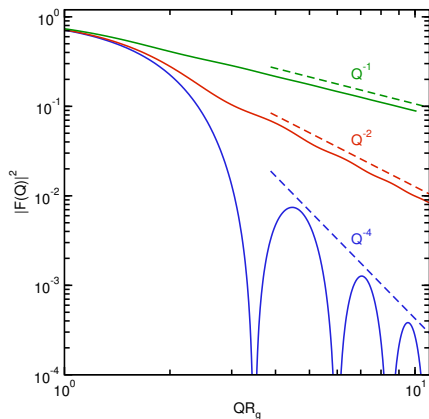
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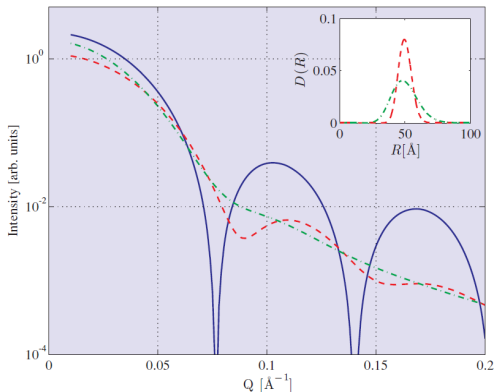
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$p = 0$

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$p = 20\%$



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- Nucleation and growth of & glycine crystals

SAXS of irradiated Zn nanoparticles

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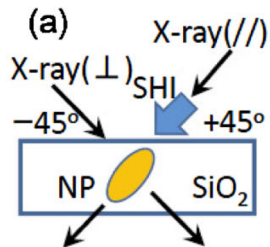
SAXS is measured using 18 keV x-rays both parallel and perpendicular to the direction of Xe⁺¹⁴ irradiation.

"Shape elongation of embedded Zn nanoparticles induced by swift heavy ion irradiation: A SAXS study", H. Amekura, K. Kono, N. Okubo, and N. Ishikawa, *Phys. Status Solidi B* **252**, 165-169 (2015).

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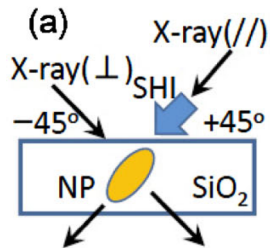
Expt. geometry

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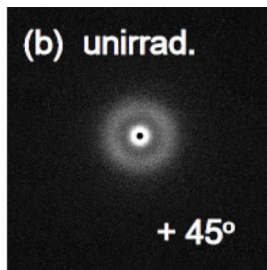
SAXS of irradiated Zn nanoparticles

Zn nanoparticles formed in SiO_2 by ion implantation are irradiated with high energy Xe^{+14} ions.

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Expt. geometry



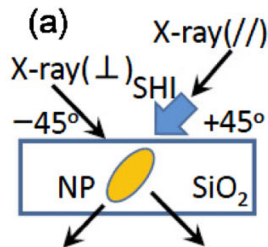
Unirradiated

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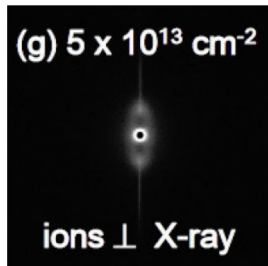
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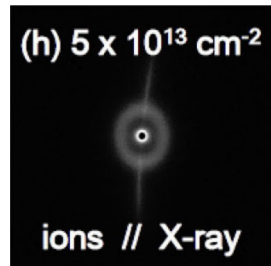
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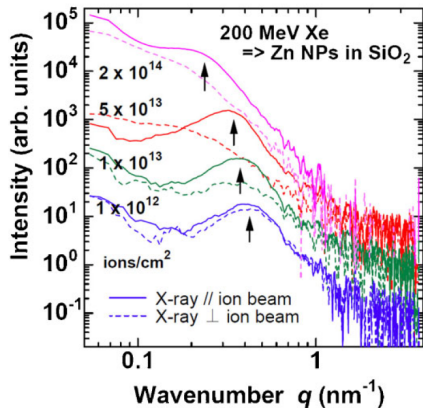
Irradiated || x-rays



Irradiated ⊥ x-rays

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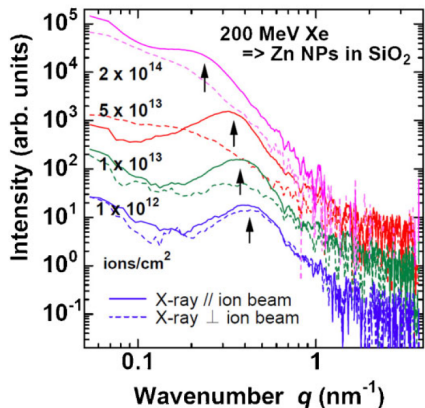
SAXS of irradiated Zn nanoparticles



SAXS intensity for // and ⊥ x-ray incidence

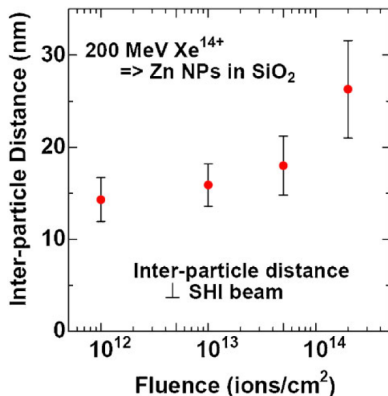
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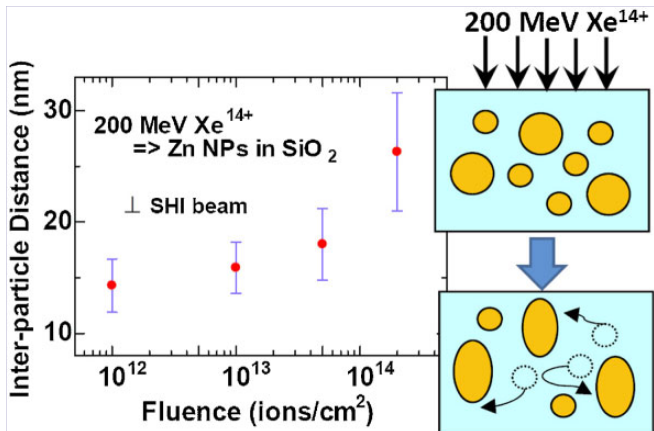
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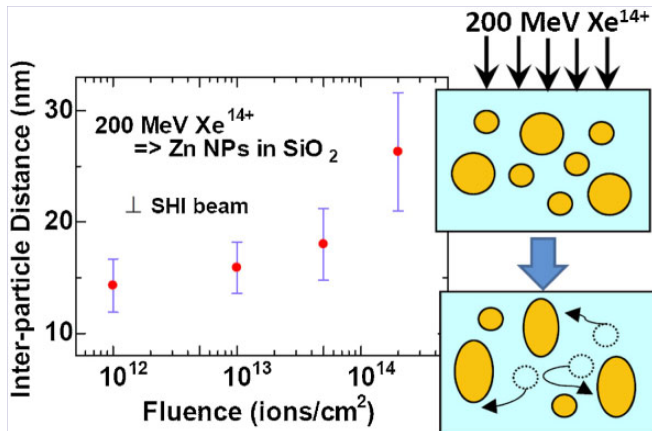
Interparticle distance as a function of irradiation fluence

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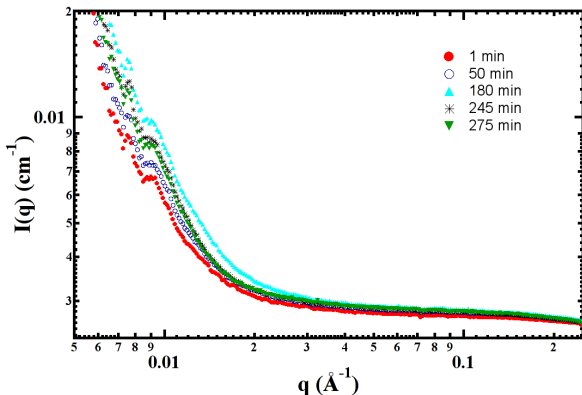


growth of interparticle spacing is due to dissolution and re-agglomeration with fluence

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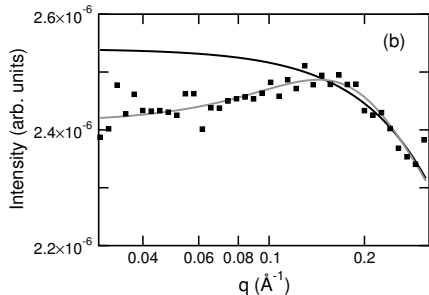
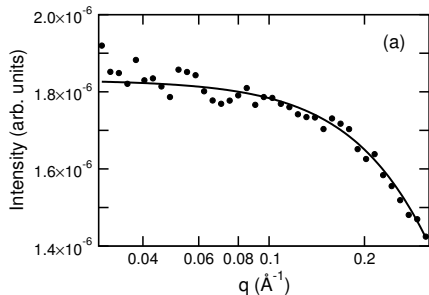
Nucleation & growth of glycine

Can SAXS help us understand the nucleation and growth of a simple molecule which is the prototype for pharmaceutical compounds?

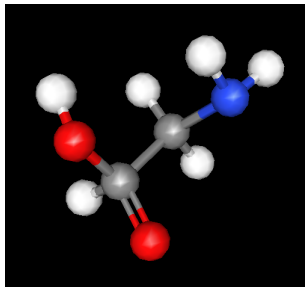


initial studies at 12 keV show change but no crystallization

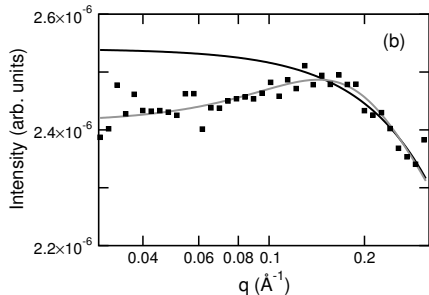
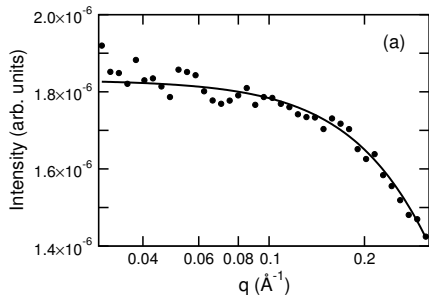
Glycine nucleation



change to 25 keV x-rays
study neutral (top) and acidic (bottom) solutions

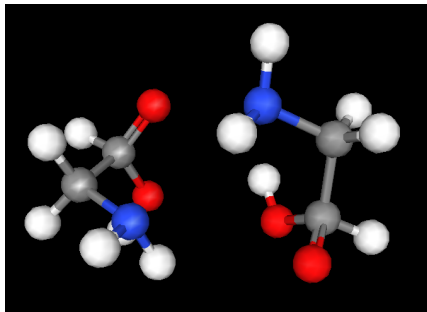


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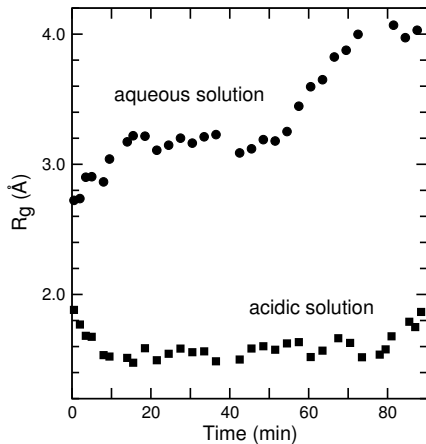


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Glycine R_g



in aqueous solution, R_g implies dimerization and increases due to aggregation until crystallization

in acidic solution, R_g remains small and implies that no dimerization or aggregation occurs before nucleation

"Relationship between Self-Association of Glycine Molecules in Supersaturated Solution and Solid State Outcome",
D. Erdemir et al. *Phys. Rev. Lett.* **99**, 115702 (2007)

Size exclusion chromatography SAXS

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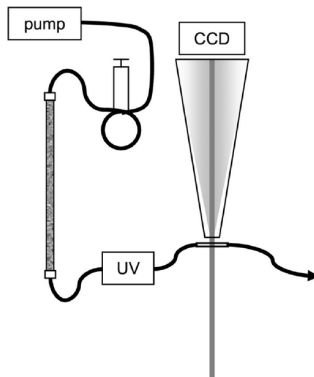
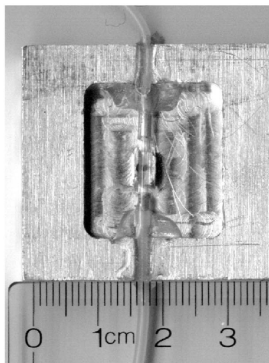
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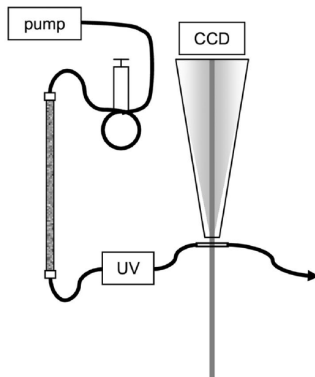
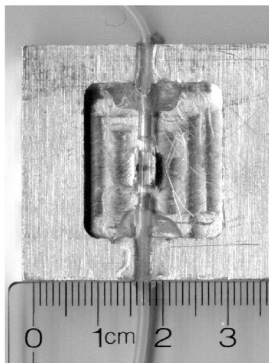
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Matthew, Mirza & Menhart, "liquid-chromatography-coupled SAXS for accurate sizing of aggregating proteins," *J. Synchrotron Rad.* **11**, 314-318 (2004) developed a technique which is now being used routinely in biological SAXS, called Size Exclusion Chromatography SAXS.

Size exclusion chromatography SAXS

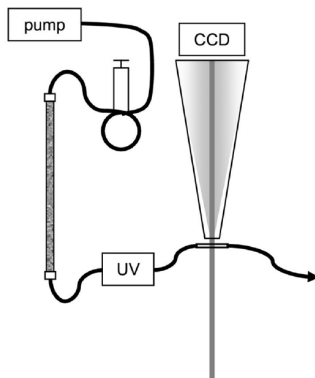
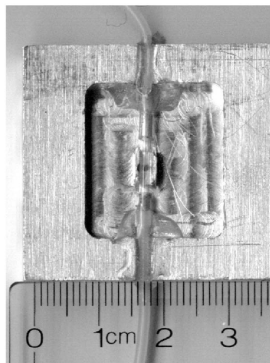


Size exclusion chromatography SAXS



2m SAXS camera, 1.03\AA (12 keV) x-rays were used

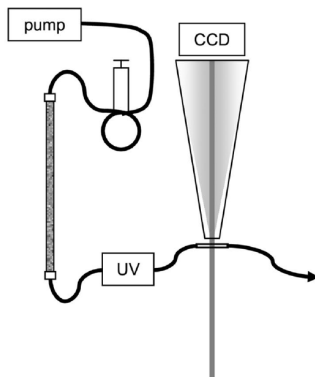
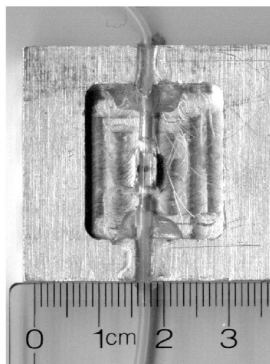
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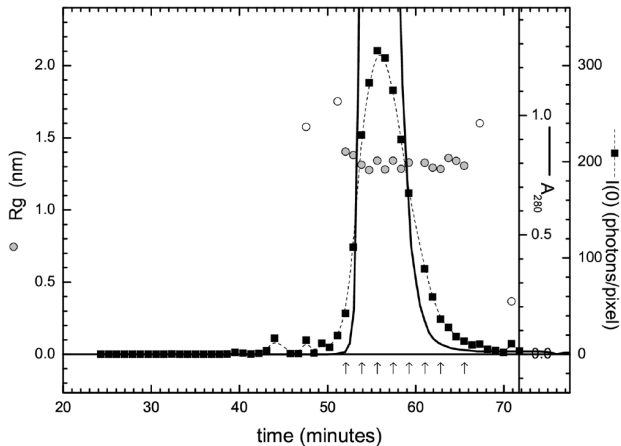


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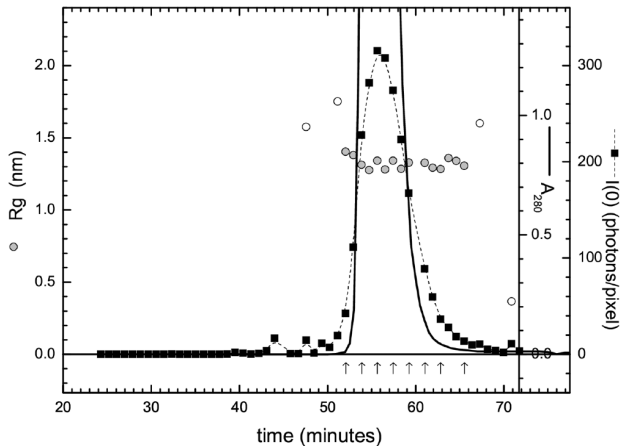
samples of (1) cytochrome c, (2) plasminogen, (3) mixture of cytochrome c bovine serum albumin, and blue dextran

SEC-SAXS experimental setup



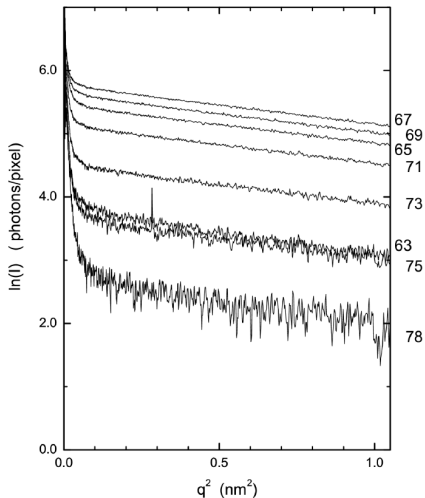
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Cytochrome c



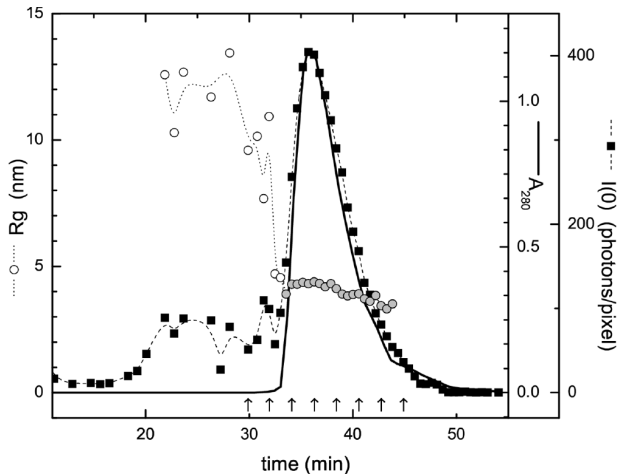
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Cytochrome c - Guinier plots



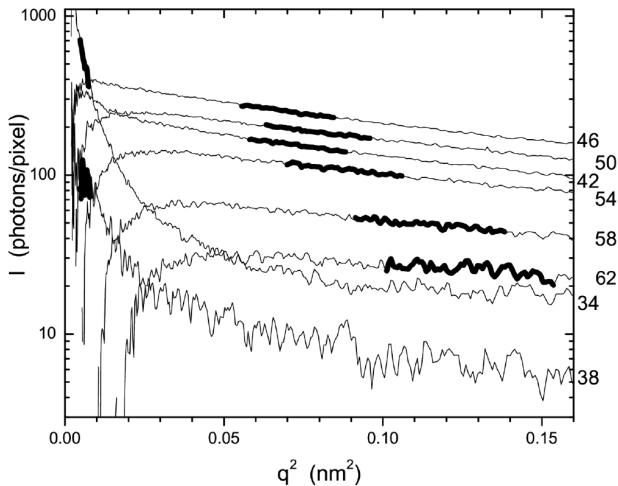
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Plasminogen



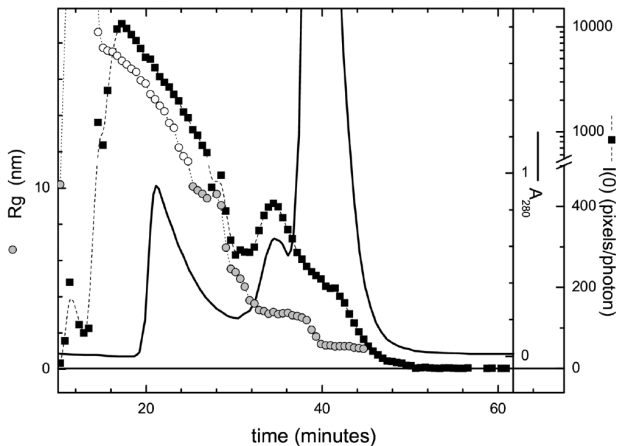
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Three component mixture

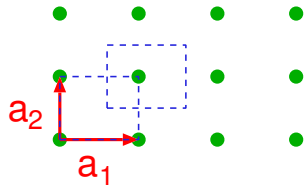


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Types of lattice vectors

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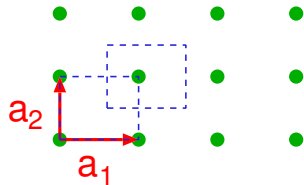
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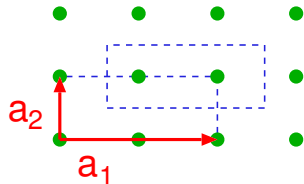
primitive

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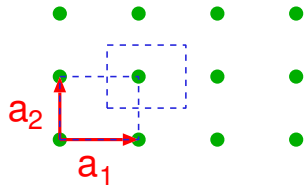
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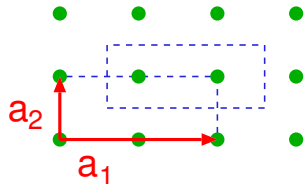
non-primitive

Types of lattice vectors

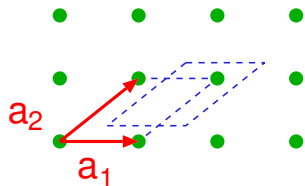


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primitive

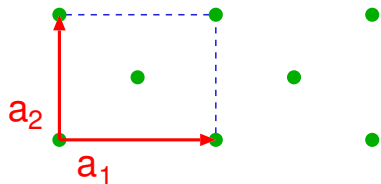


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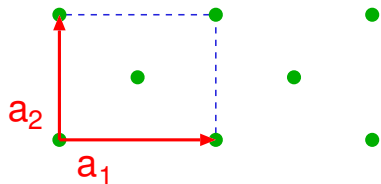
non-conventional

More about lattice vectors

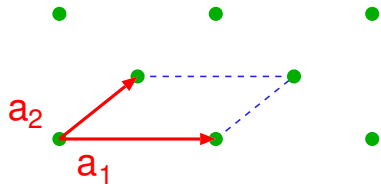


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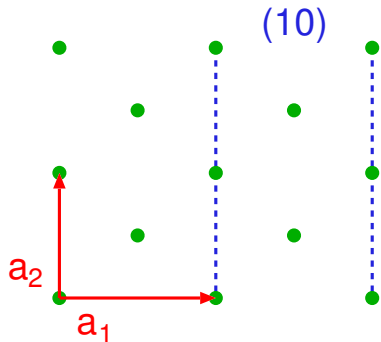


sometimes conventional axes...



...are not primitive

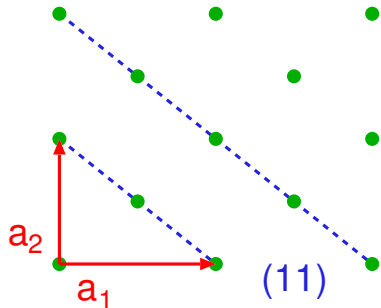
Miller indices



planes designated (hk) , intercept the unit cell axes at

$$\frac{a_1}{h}, \quad \frac{a_2}{k}$$

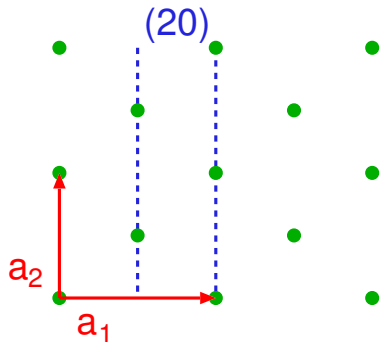
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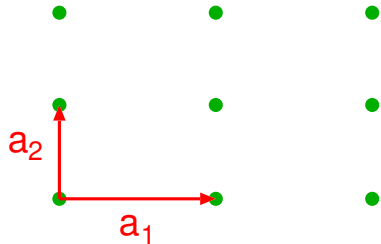
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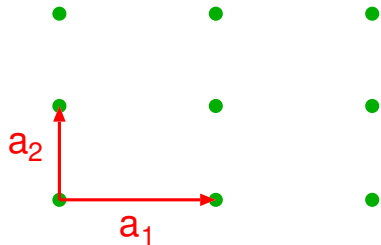
for a lattice with orthogonal unit vectors

$$\frac{1}{d_{hk}^2} = \frac{h^2}{a_1^2} + \frac{k^2}{a_2^2}$$

Reciprocal lattice



Reciprocal lattice

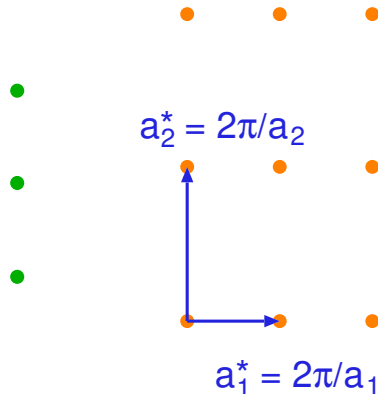
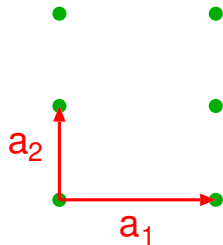


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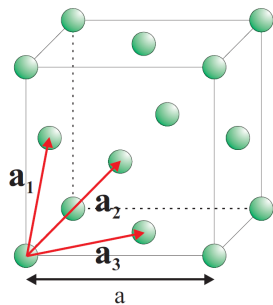
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$$\therefore \vec{Q} = \vec{G}_{hkl}$$

The FCC reciprocal lattice

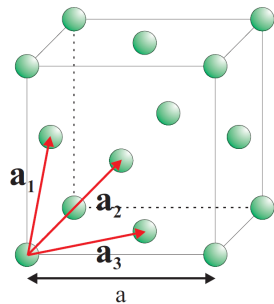
The primitive lattice vectors of the face-centered cubic lattice are



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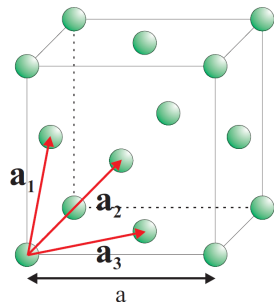
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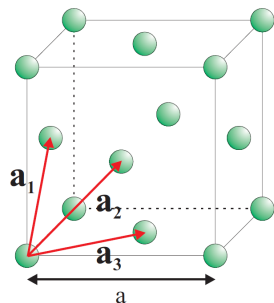
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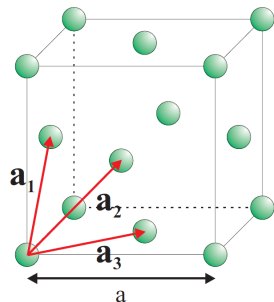


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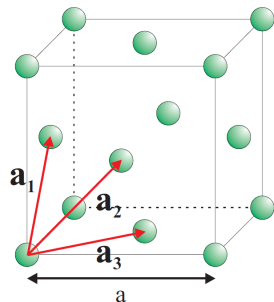
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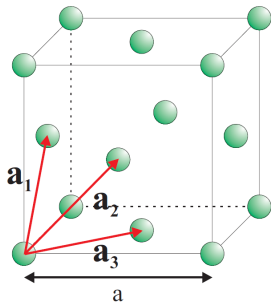
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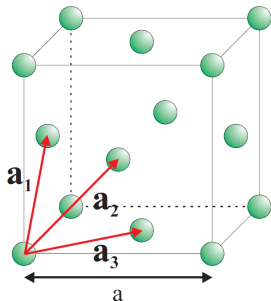
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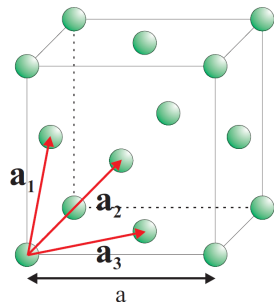
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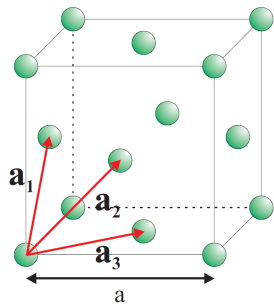
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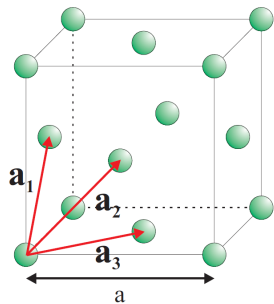
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$$\vec{a}_1 = \frac{a}{2}(\hat{y} + \hat{z}), \quad \vec{a}_2 = \frac{a}{2}(\hat{z} + \hat{x}), \quad \vec{a}_3 = \frac{a}{2}(\hat{x} + \hat{y})$$



The volume of the unit cell is

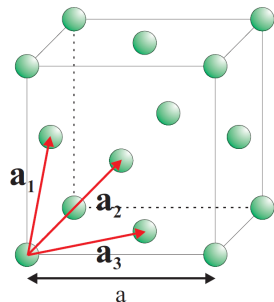
$$v_c = \vec{a}_1 \cdot \vec{a}_2 \times \vec{a}_3 = \vec{a}_1 \cdot \frac{a^2}{4}(\hat{y} + \hat{z} - \hat{x}) = \frac{a^3}{4}$$

$$\begin{aligned} \vec{a}_1^* &= \frac{2\pi}{v_c} \vec{a}_2 \times \vec{a}_3 = \frac{2\pi}{v_c} \frac{a^2}{4}(\hat{y} + \hat{z} - \hat{x}) \\ &= \frac{4\pi}{a} \left(\frac{\hat{y}}{2} + \frac{\hat{z}}{2} - \frac{\hat{x}}{2} \right) \end{aligned}$$

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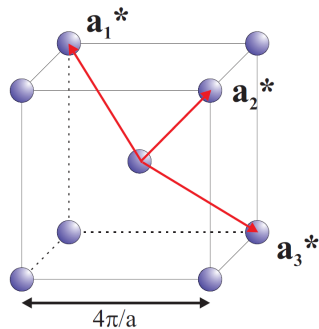
$$\vec{a}_2^* = \frac{4\pi}{a} \left(\frac{\hat{z}}{2} + \frac{\hat{x}}{2} - \frac{\hat{y}}{2} \right)$$

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which is a body-centered cubic lattice

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