## Today's Outline - October 03, 2016

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- Scattering from two electrons


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- Scattering from atoms
- Scattering from molecules


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- Scattering from two electrons
- Scattering from atoms
- Scattering from molecules
- Radial distribution function


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APS Visits:
10-ID: Friday, October 21, 2016
10-BM: Friday, October 28, 2016
Homework Assignment \#03:
Chapter 3: 1, 3, 4, 6, 8
due Wednesday, September 05, 2016

## Scattering from two electrons

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Since experiments measure $I \propto A^{2}$, the phase information is lost. This is a problem if we don't know the specific orientation of the scattering system relative to the x-ray beam.

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We will now look at the consequences of this orientation and generalize to more than two electrons

## Two electrons - fixed orientation

The expression

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I(\vec{Q})=2 r_{0}^{2}(1+\cos (\vec{Q} \cdot \vec{r}))
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assumes that the two electrons have a specific, fixed orientation. In this case the intensity as a function of $Q$ is.

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Fixed orientation is not the usual case, particularly for solution and small-angle scattering.


## Orientation averaging

Consider scattering from two arbitrary electron distributions, $f_{1}$ and $f_{2}$. $A(\vec{Q})$, is given by

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A(\vec{Q})=f_{1}+f_{2} e^{i \vec{Q} \cdot \vec{F}}
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## Randomly oriented electrons

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When we now replace the two arbitrary scattering distributions with electrons $\left(f_{1}, f_{2} \rightarrow-r_{0}\right)$, we change the intensity profile significantly.


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Single electrons are a good first example but a real system involves scattering from atoms. We can use what we have already used to write an expression for the scattering from an atom, ignoring the anomalous terms.

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Z_{H e}=2 \quad Z_{A r}=18
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From the atomic form factor, we would like to abstract to the next level of complexity, a molecule (we will leave crystals for Chapter 5).

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\left|F^{\text {mol }}\right|^{2}=\left|f^{C}\right|^{2}+4\left|f^{F}\right|^{2}+8 f^{C} f^{F} \frac{\sin (Q R)}{Q R}+12\left|f^{F}\right|^{2} \frac{\sin (Q \sqrt{8 / 3} R)}{Q \sqrt{8 / 3} R}
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## The Radial Distribution Function

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Ordered 2D crystal
Amorphous solid or liquid

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