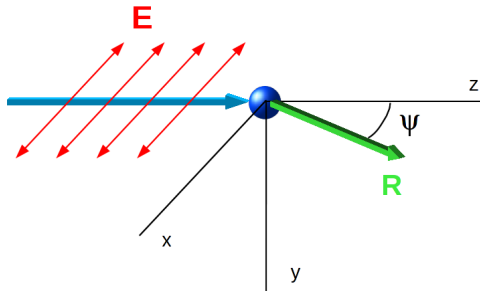


Review: Thomson Scattering



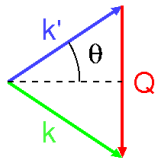
$$\frac{E_{rad}(R, t)}{E_{in}} = -\frac{e^2}{4\pi\epsilon_0 mc^2} \frac{e^{ikR}}{R} \cos\psi = -r_o \frac{e^{ikR}}{R} \cos\psi$$

$$r_o = \frac{e^2}{4\pi\epsilon_0 mc^2} = 2.82 \times 10^{-5} \text{\AA}$$

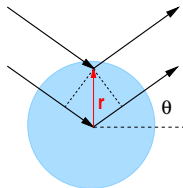
Review: Atomic Scattering

phase shift arises from scattering off different portions of extended electron distribution

$$\Delta\phi(\mathbf{r}) = (\mathbf{k} - \mathbf{k}') \cdot \mathbf{r} = \mathbf{Q} \cdot \mathbf{r}$$



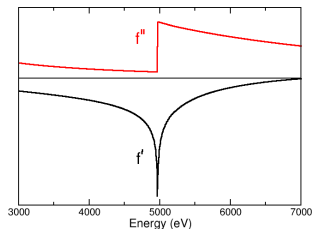
$$|\mathbf{Q}| = 2 |\mathbf{k}| \sin\theta = \frac{4\pi}{\lambda} \sin\theta$$



Review: Atomic Form Factor

the volume element at $\mathbf{r}$ contributes $-r_o\rho(\mathbf{r})d^3r$ with phase factor $e^{i\mathbf{Q}\cdot\mathbf{r}}$ for an entire atom, integrate to get the atomic form factor $f^o(\mathbf{Q})$:

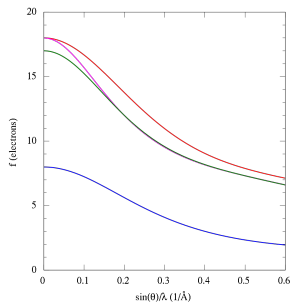
$$-r_o f^o(\mathbf{Q}) = -r_o \int \rho(\mathbf{r}) e^{i\mathbf{Q}\cdot\mathbf{r}} d^3r$$



the total atomic scattering factor is

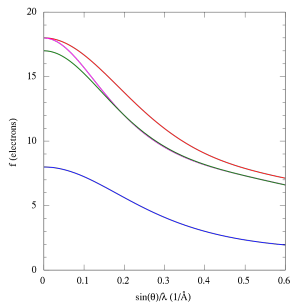
$$f(\mathbf{Q}, \hbar\omega) = f^o(\mathbf{Q}) + f'(\hbar\omega) + if''(\hbar\omega)$$

Atomic Form Factor



The atomic form factor has an angular dependence

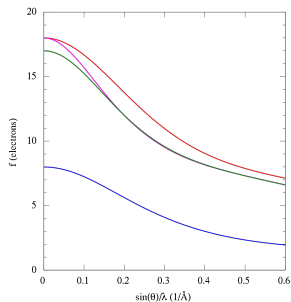
Atomic Form Factor



The atomic form factor has an angular dependence

$$Q = \frac{4\pi}{\lambda} \sin \theta$$

Atomic Form Factor



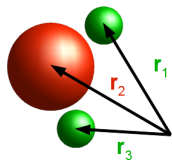
The atomic form factor has an angular dependence

$$Q = \frac{4\pi}{\lambda} \sin \theta$$

Lighter atoms (blue is oxygen) have wider form factor

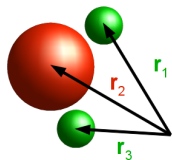
Scattering From Molecules

extending to a molecule ...



Scattering From Molecules

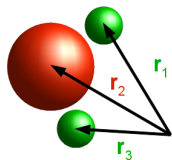
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$$F^{molecule}(\mathbf{Q}) = \sum_j f_j(\mathbf{Q}) e^{i\mathbf{Q} \cdot \mathbf{r}_j}$$

Scattering From Molecules

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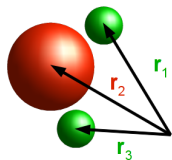


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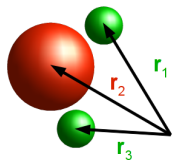


$$F^{molecule}(\mathbf{Q}) = \sum_j f_j(\mathbf{Q}) e^{i\mathbf{Q} \cdot \mathbf{r}_j}$$

$$F^{molecule}(\mathbf{Q}) = f_1(\mathbf{Q}) e^{i\mathbf{Q} \cdot \mathbf{r}_1} +$$

Scattering From Molecules

extending to a molecule ...

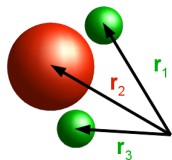


$$F^{molecule}(\mathbf{Q}) = \sum_j f_j(\mathbf{Q}) e^{i\mathbf{Q} \cdot \mathbf{r}_j}$$

$$F^{molecule}(\mathbf{Q}) = f_1(\mathbf{Q}) e^{i\mathbf{Q} \cdot \mathbf{r}_1} + f_2(\mathbf{Q}) e^{i\mathbf{Q} \cdot \mathbf{r}_2} +$$

Scattering From Molecules

extending to a molecule ...



$$F^{molecule}(\mathbf{Q}) = \sum_j f_j(\mathbf{Q}) e^{i\mathbf{Q} \cdot \mathbf{r}_j}$$

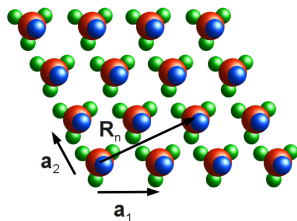
$$F^{molecule}(\mathbf{Q}) = f_1(\mathbf{Q}) e^{i\mathbf{Q} \cdot \mathbf{r}_1} + f_2(\mathbf{Q}) e^{i\mathbf{Q} \cdot \mathbf{r}_2} + f_3(\mathbf{Q}) e^{i\mathbf{Q} \cdot \mathbf{r}_3}$$

Scattering from a Crystal

The scattering from a molecule can be extended to a crystal lattice ...

Scattering from a Crystal

The scattering from a molecule can be extended to a crystal lattice ...

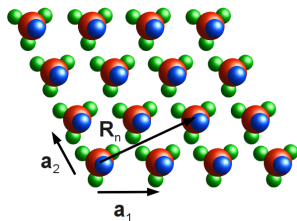


... which is simply a periodic array of molecules

$$F^{crystal}(\mathbf{Q}) = F^{molecule} F^{lattice}$$

Scattering from a Crystal

The scattering from a molecule can be extended to a crystal lattice ...



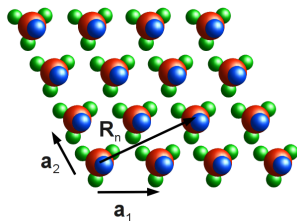
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$$F^{crystal}(\mathbf{Q}) = \sum_j f_j(\mathbf{Q}) e^{i\mathbf{Q} \cdot \mathbf{r}_j} \sum_n e^{i\mathbf{Q} \cdot \mathbf{R}_n}$$

Scattering from a Crystal

The scattering from a molecule can be extended to a crystal lattice ...



... which is simply a periodic array of molecules

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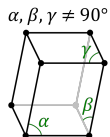
$\sum e^{i\mathbf{Q} \cdot \mathbf{R}_n}$ is always small

unless $\mathbf{Q} \cdot \mathbf{R}_n = 2\pi m$

where $\mathbf{R}_n = n_1 \mathbf{a}_1 + n_2 \mathbf{a}_2 + n_3 \mathbf{a}_3$ is a real space lattice vector

Crystal Lattices

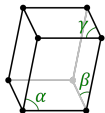
There are 7 possible real space lattices: triclinic,



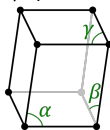
Crystal Lattices

There are 7 possible real space lattices: triclinic, monoclinic,

$$\alpha, \beta, \gamma \neq 90^\circ$$



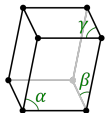
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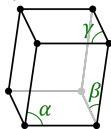
Crystal Lattices

There are 7 possible real space lattices: triclinic, monoclinic, orthorhombic,

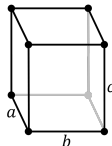
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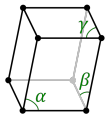
$$a \neq b \neq c$$



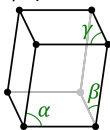
Crystal Lattices

There are 7 possible real space lattices: triclinic, monoclinic, orthorhombic, tetragonal,

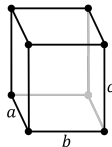
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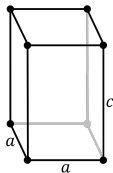
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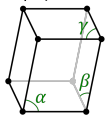
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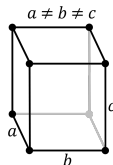
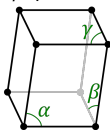
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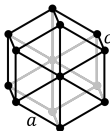
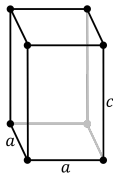
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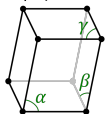
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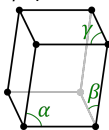
Crystal Lattices

There are 7 possible real space lattices: triclinic, monoclinic, orthorhombic, tetragonal, hexagonal, rhombohedral,

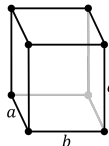
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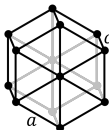
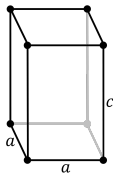
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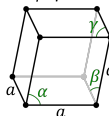
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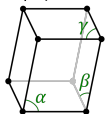
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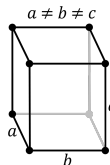
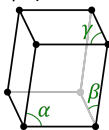
Crystal Lattices

There are 7 possible real space lattices: triclinic, monoclinic, orthorhombic, tetragonal, hexagonal, rhombohedral, cubic

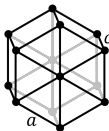
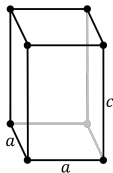
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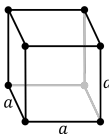
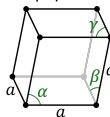
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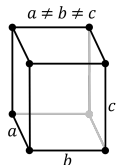


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Lattice Volume

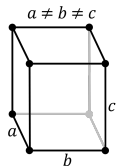
Consider the orthorhombic lattice for simplicity (the others give exactly the same result).



Lattice Volume

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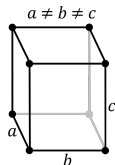
$$\mathbf{a}_1 = a\hat{\mathbf{x}}, \quad \mathbf{a}_2 = b\hat{\mathbf{y}}, \quad \mathbf{a}_3 = c\hat{\mathbf{z}}$$



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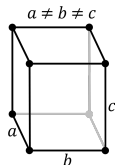


$$\mathbf{a}_1 \times \mathbf{a}_2 = ab\hat{\mathbf{z}}$$

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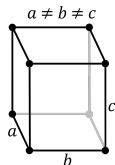
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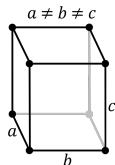
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$$(\mathbf{a}_1 \times \mathbf{a}_2) \cdot \mathbf{a}_3 = abc = V$$

A simple way of calculating the volume of the unit cell!

Reciprocal Lattice

Define the reciprocal lattice vectors in terms of the real space unit vectors

$$\mathbf{a}_1^* = 2\pi \frac{\mathbf{a}_2 \times \mathbf{a}_3}{\mathbf{a}_1 \cdot (\mathbf{a}_2 \times \mathbf{a}_3)}$$

$$\mathbf{a}_2^* = 2\pi \frac{\mathbf{a}_3 \times \mathbf{a}_1}{\mathbf{a}_2 \cdot (\mathbf{a}_3 \times \mathbf{a}_1)}$$

$$\mathbf{a}_3^* = 2\pi \frac{\mathbf{a}_1 \times \mathbf{a}_2}{\mathbf{a}_3 \cdot (\mathbf{a}_1 \times \mathbf{a}_2)}$$

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$$\mathbf{a}_3^* = 2\pi \frac{\mathbf{a}_1 \times \mathbf{a}_2}{\mathbf{a}_3 \cdot (\mathbf{a}_1 \times \mathbf{a}_2)}$$

In analogy to $\mathbf{R}_n$, we can construct an arbitrary reciprocal space lattice vector $\mathbf{G}_{hkl}$

$$\mathbf{G}_{hkl} = h\mathbf{a}_1^* + k\mathbf{a}_2^* + l\mathbf{a}_3^*$$

Laue Condition

Because of the construction of the reciprocal lattice

$$\mathbf{G}_{hkl} \cdot \mathbf{R}_n = 2\pi(hn_1 + kn_2 + ln_3) = 2\pi m$$

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and therefore, the crystal scattering factor is non-zero **only** when

$$\mathbf{Q} = \mathbf{G}_{hkl}$$

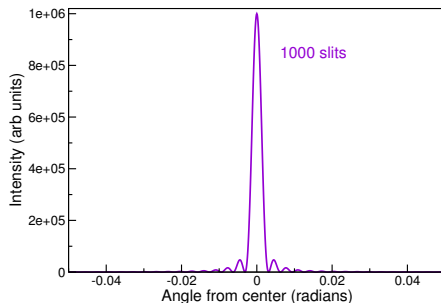
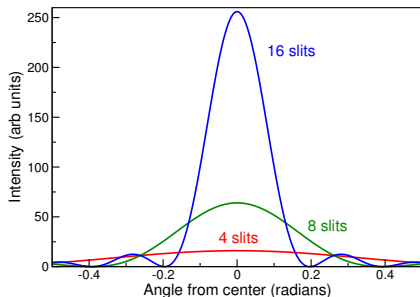
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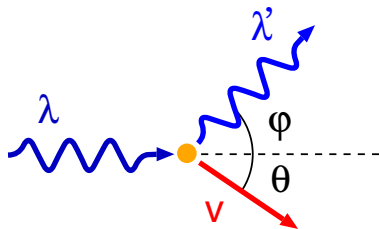
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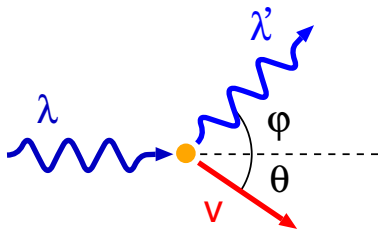
Compton Scattering

A photon-electron collision



Compton Scattering

A photon-electron collision



$$\mathbf{p} = \hbar \mathbf{k} = 2\pi \hbar / \lambda$$

$$\mathbf{p}' = \hbar \mathbf{k}' = 2\pi \hbar / \lambda'$$

$$|\mathbf{k}| \neq |\mathbf{k}'|$$

Treat electron relativistically and conserve energy and momentum

$$mc^2 + \frac{hc}{\lambda} = \frac{hc}{\lambda'} + \gamma mc^2 \quad (\text{energy})$$

$$\frac{h}{\lambda} = \frac{h}{\lambda'} \cos \phi + \gamma mv \cos \theta \quad (\text{x-axis})$$

$$0 = \frac{h}{\lambda'} \sin \phi + \gamma mv \sin \theta \quad (\text{y-axis})$$

Compton Scattering Derivation

Squaring the momentum equations

$$\left(\frac{h}{\lambda} - \frac{h}{\lambda'} \cos \phi \right)^2 = \gamma^2 m^2 v^2 \cos^2 \theta$$

Compton Scattering Derivation

Squaring the momentum equations

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using $\sin^2 \theta + \cos^2 \theta = 1$

$$\left(\frac{h}{\lambda} - \frac{h}{\lambda'} \cos \phi \right)^2 = \gamma^2 m^2 v^2 \left[1 - \frac{\left(\frac{h}{\lambda'} \sin \phi \right)^2}{\gamma^2 m^2 v^2} \right]$$

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Squaring the momentum equations

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$$\frac{h^2}{\lambda^2} - \frac{2h^2}{\lambda\lambda'} \cos \phi + \frac{h^2}{\lambda'^2} \cos^2 \phi = \gamma^2 m^2 v^2 - \frac{h^2}{\lambda'^2} \sin^2 \phi$$

Compton Scattering Derivation

Squaring the momentum equations

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$$\frac{h^2}{\lambda^2} - \frac{2h^2}{\lambda\lambda'} \cos \phi + \frac{h^2}{\lambda'^2} = \frac{m^2 v^2}{1-\beta^2} = \frac{m^2 c^2 \beta^2}{1-\beta^2}$$

Compton Scattering Derivation

Now take the energy equation and square it

$$\left(mc^2 + \frac{hc}{\lambda} - \frac{hc}{\lambda'} \right)^2 = \gamma^2 m^2 c^4 = \frac{m^2 c^4}{1 - \beta^2}$$

Compton Scattering Derivation

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$$\left(mc^2 + \frac{hc}{\lambda} - \frac{hc}{\lambda'} \right)^2 - \beta^2 \left(mc^2 + \frac{hc}{\lambda} - \frac{hc}{\lambda'} \right)^2 = m^2 c^4$$

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$$\beta^2 = \frac{\left(mc^2 + \frac{hc}{\lambda} - \frac{hc}{\lambda'} \right)^2 - m^2 c^4}{\left(mc^2 + \frac{hc}{\lambda} - \frac{hc}{\lambda'} \right)^2} = \frac{2mc^2 \left(\frac{hc}{\lambda} - \frac{hc}{\lambda'} \right) + \left(\frac{hc}{\lambda} - \frac{hc}{\lambda'} \right)^2}{\left(mc^2 + \frac{hc}{\lambda} - \frac{hc}{\lambda'} \right)^2}$$

Compton Scattering Derivation

Combining and reducing, we obtain

$$\frac{h^2}{\lambda^2} + \frac{h^2}{\lambda'^2} - \frac{2h^2}{\lambda\lambda'} \cos \phi = 2m \left(\frac{hc}{\lambda} - \frac{hc}{\lambda'} \right) + \frac{h^2}{\lambda^2} + \frac{h^2}{\lambda'^2} - \frac{2h^2}{\lambda\lambda'}$$

Compton Scattering Derivation

Combining and reducing, we obtain

$$\frac{h^2}{\lambda^2} + \frac{h^2}{\lambda'^2} - \frac{2h^2}{\lambda\lambda'} \cos \phi = 2m \left(\frac{hc}{\lambda} - \frac{hc}{\lambda'} \right) + \frac{h^2}{\lambda^2} + \frac{h^2}{\lambda'^2} - \frac{2h^2}{\lambda\lambda'}$$
$$\frac{2h^2}{\lambda\lambda'} (1 - \cos \phi) = 2m \left(\frac{hc}{\lambda} - \frac{hc}{\lambda'} \right) = 2mhc \left(\frac{\lambda' - \lambda}{\lambda\lambda'} \right) = \frac{2mhc\Delta\lambda}{\lambda\lambda'}$$

Compton Scattering Derivation

Combining and reducing, we obtain

$$\begin{aligned}\frac{h^2}{\lambda^2} + \frac{h^2}{\lambda'^2} - \frac{2h^2}{\lambda\lambda'} \cos \phi &= 2m \left(\frac{hc}{\lambda} - \frac{hc}{\lambda'} \right) + \frac{h^2}{\lambda^2} + \frac{h^2}{\lambda'^2} - \frac{2h^2}{\lambda\lambda'} \\ \frac{2h^2}{\lambda\lambda'} (1 - \cos \phi) &= 2m \left(\frac{hc}{\lambda} - \frac{hc}{\lambda'} \right) = 2mhc \left(\frac{\lambda' - \lambda}{\lambda\lambda'} \right) = \frac{2mhc\Delta\lambda}{\lambda\lambda'} \\ \Delta\lambda &= \frac{h}{mc} (1 - \cos \phi)\end{aligned}$$

Compton Scattering Results

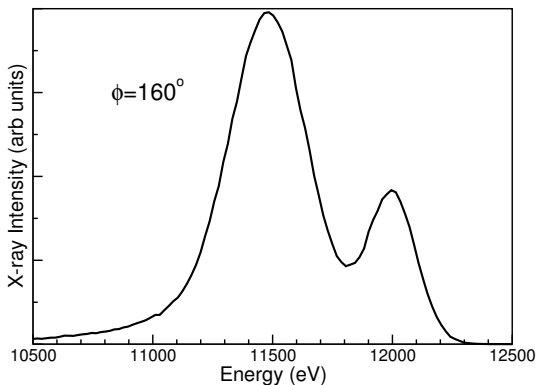
$$\lambda_c = \hbar/mc = 3.86 \times 10^{-3} \text{\AA} \text{ for an electron}$$

Comparing to the Thomson scattering length: $r_o/\lambda_C = 1/137$

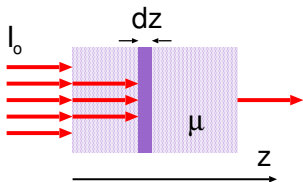
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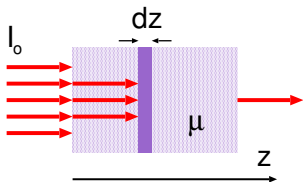
X-Ray Absorption



Absorption coefficient μ , thickness dz
x-ray intensity is attenuated as

$$dI = -I(z)\mu dz$$

X-Ray Absorption

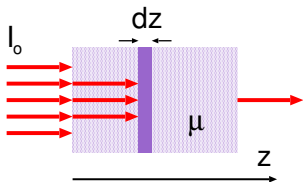


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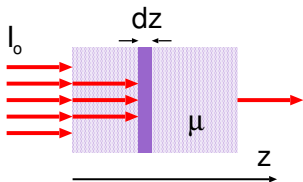
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where ρ_a is atom density, σ_a is absorption cross section

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where ρ_a is atom density, σ_a is absorption cross section

$$\mu = \rho_a\sigma_a = \left(\frac{\rho_m N_A}{A} \right) \sigma_a$$

with mass density ρ_m , Avogadro's number N_A , atomic number A