

The statistical interpretation





- The meaning of the wavefunction

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- Probability review

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- Expectation values

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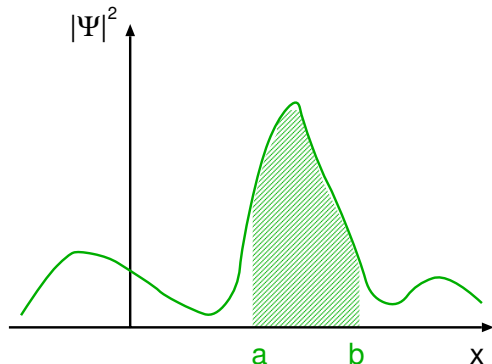
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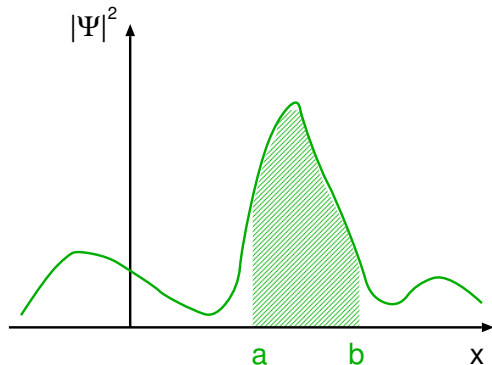
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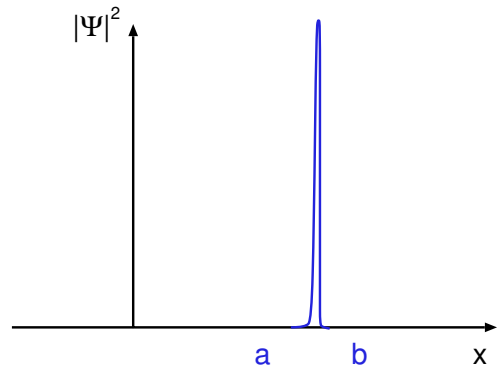
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the Copenhagen interpretation has proven to be correct in that it accurately describes experimental measurements

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Since $\sigma^2 \geq 0$, $\langle j^2 \rangle \geq \langle j \rangle^2$

Wavefunction normalization





- Normalizing the wavefunction



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- Time invariance of normalization

Continuous variables



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Normalizing the wave function



Beyond satisfying the Schrödinger equation, the wave function must also have physical significance.

$$i\hbar \frac{\partial \Psi}{\partial t} = -\frac{\hbar^2}{2m} \frac{\partial^2 \Psi}{\partial x^2} + V\Psi$$

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Beyond satisfying the Schrödinger equation, the wave function must also have physical significance. If we are to believe the statistical interpretation, the wave-function must be *normalized*, that is the integral of the probability density over all space must be unity.

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Time independence of normalization



Look more closely at the integrand and apply the product rule.

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$$\begin{aligned}\frac{\partial}{\partial t} |\Psi|^2 &= \frac{\partial}{\partial t} (\Psi^* \Psi) \\ &= \Psi^* \frac{\partial \Psi}{\partial t} + \frac{\partial \Psi^*}{\partial t} \Psi \\ &= \Psi^* \frac{i\hbar}{2m} \frac{\partial^2 \Psi}{\partial x^2} - \frac{i\hbar}{2m} \frac{\partial^2 \Psi^*}{\partial x^2} \Psi \\ &= \frac{i\hbar}{2m} \left(\Psi^* \frac{\partial^2 \Psi}{\partial x^2} - \frac{\partial^2 \Psi^*}{\partial x^2} \Psi \right)\end{aligned}$$

using the Schrödinger equation

$$\frac{\partial \Psi}{\partial t} = \frac{i\hbar}{2m} \frac{\partial^2 \Psi}{\partial x^2} - \frac{i}{\hbar} V \Psi$$

and its complex conjugate

$$\frac{\partial \Psi^*}{\partial t} = -\frac{i\hbar}{2m} \frac{\partial^2 \Psi^*}{\partial x^2} + \frac{i}{\hbar} V \Psi^*$$

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Putting the integrand back into the original expression reveals that we have an exact differential which can be immediately integrated

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Thus the wave function remains normalized at all times $t \geq 0$.





- The position operator



- The position operator
- The momentum operator



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- The uncertainty principle

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This will eventually lead directly to what is called the “bra-ket” notation commonly used in quantum mechanics.

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Momentum operator



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Uncertainty principle



Once we accept that a particle has wave nature and can be described by a normalized wave function, we can ask what is the wavelength and where is the particle's position.

Uncertainty principle



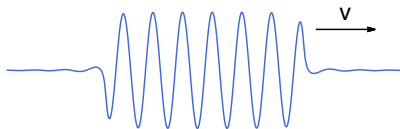
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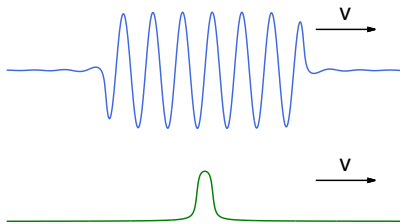
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Uncertainty principle

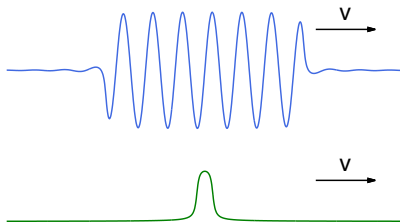


Once we accept that a particle has wave nature and can be described by a normalized wave function, we can ask what is the wavelength and where is the particle's position. Consider two cases for the wave function:

extended oscillations: a well-defined wavelength but ill-defined position

pulse-like: an ill-defined wavelength but well-defined position

The wavelength is directly related to the momentum by the de Broglie formula.



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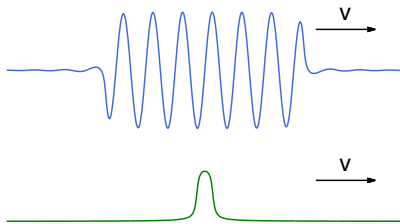


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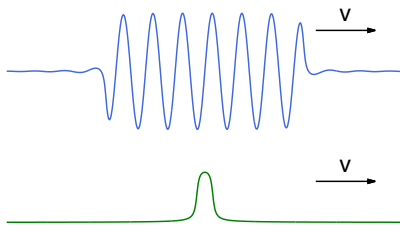
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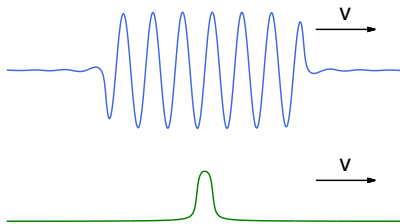
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$$p = \frac{h}{\lambda} = \frac{2\pi\hbar}{\lambda}$$
$$\sigma_x \sigma_p \geq \frac{\hbar}{2}$$

